

Computational Modeling of Perovskite Solar Cells

Edoardo Mosconi

Computational Laboratory for Hybrid/Organic
Photovoltaics - www.clhyo.org

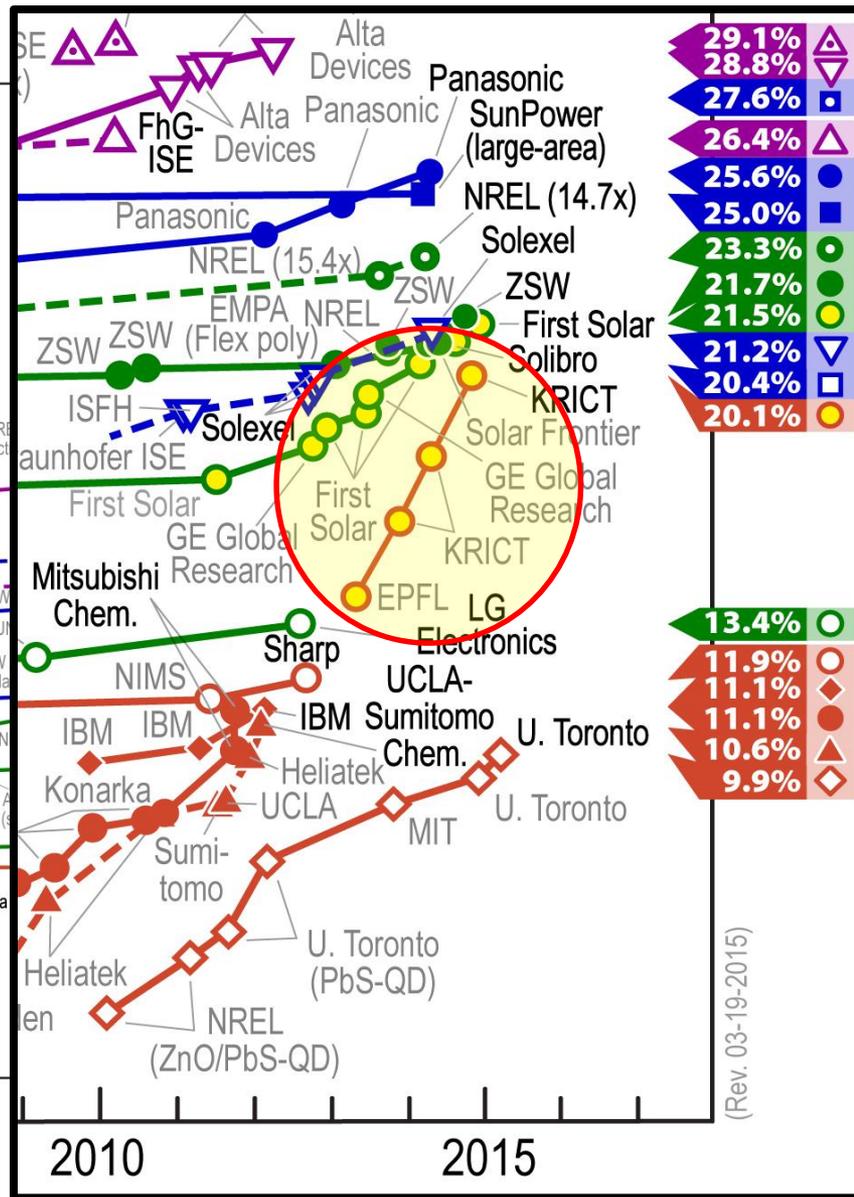
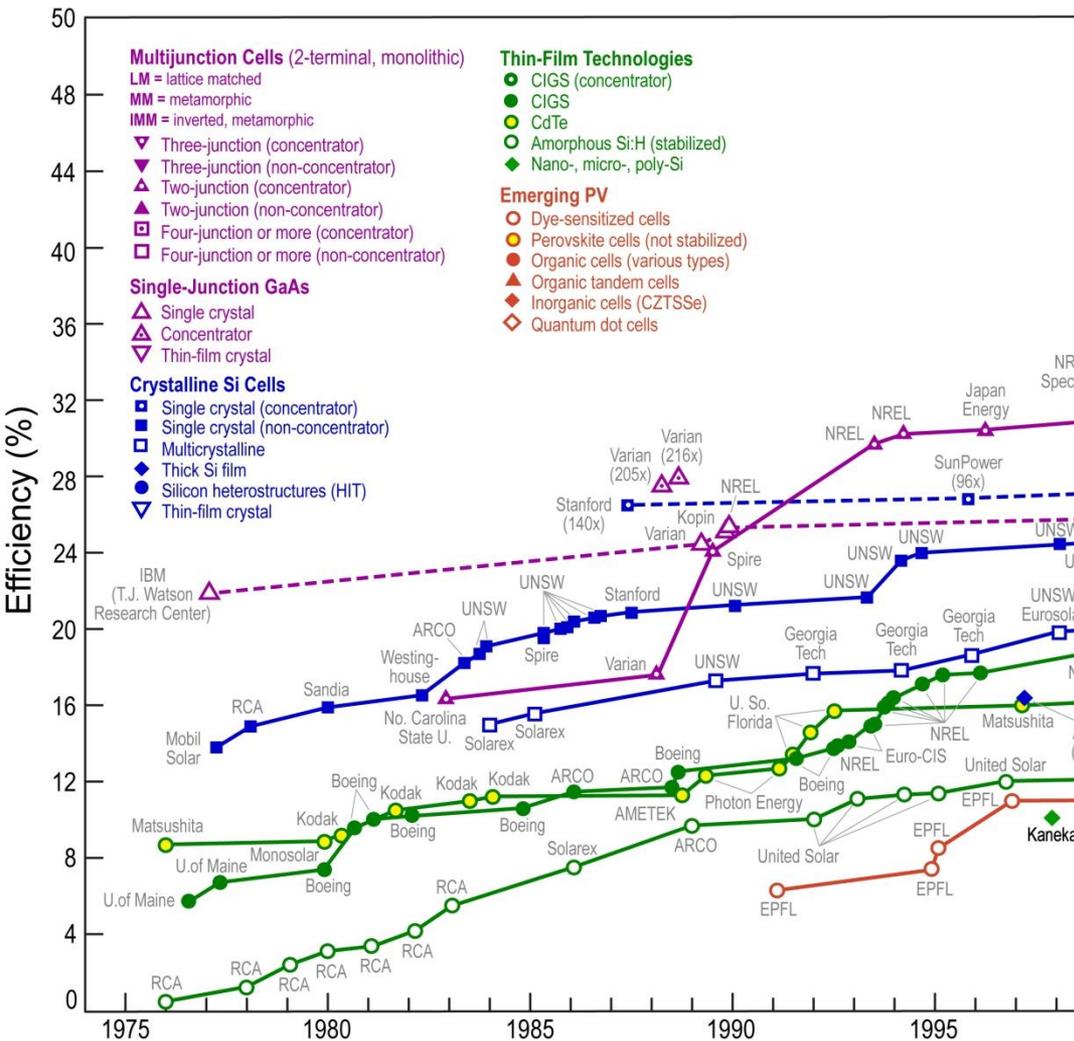
CNR- ISTM Perugia, Italy

Outline

- **Introduction**
- **Computational Approach**
- **Interfaces in PSC**
- **Mobile Ions in PSC**

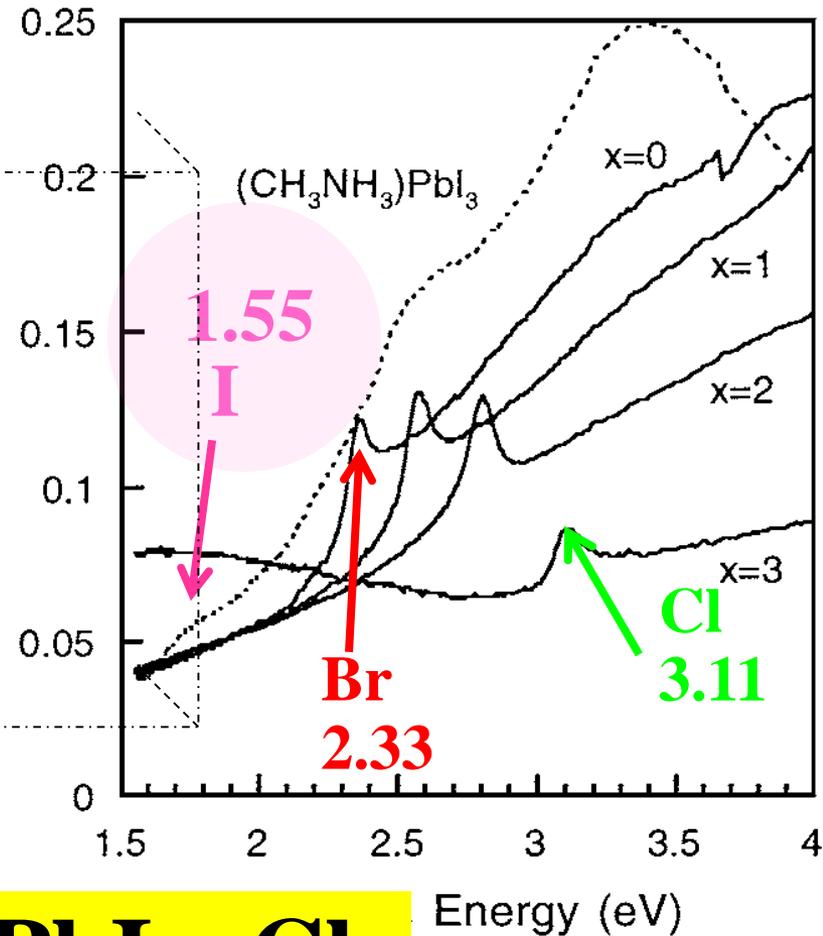
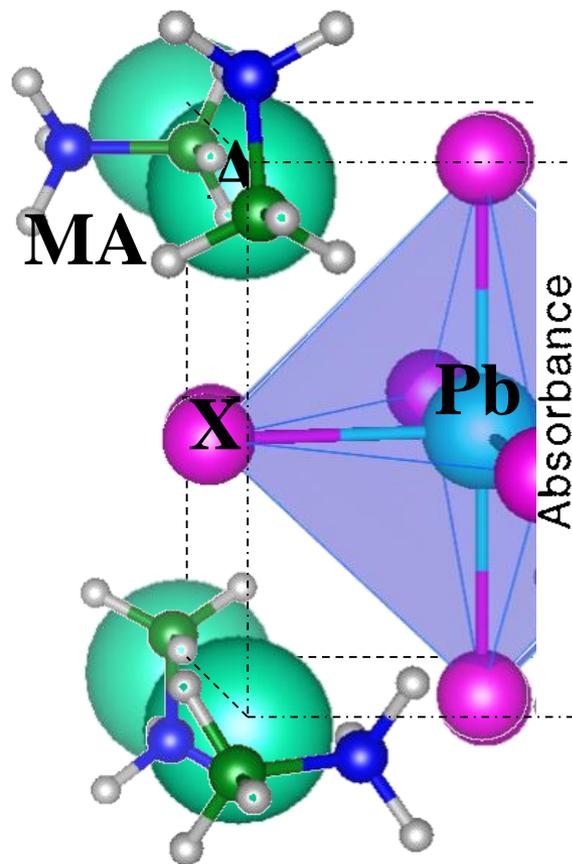
Perovskite Solar Cells Efficiency

Best Research-Cell Efficiencies



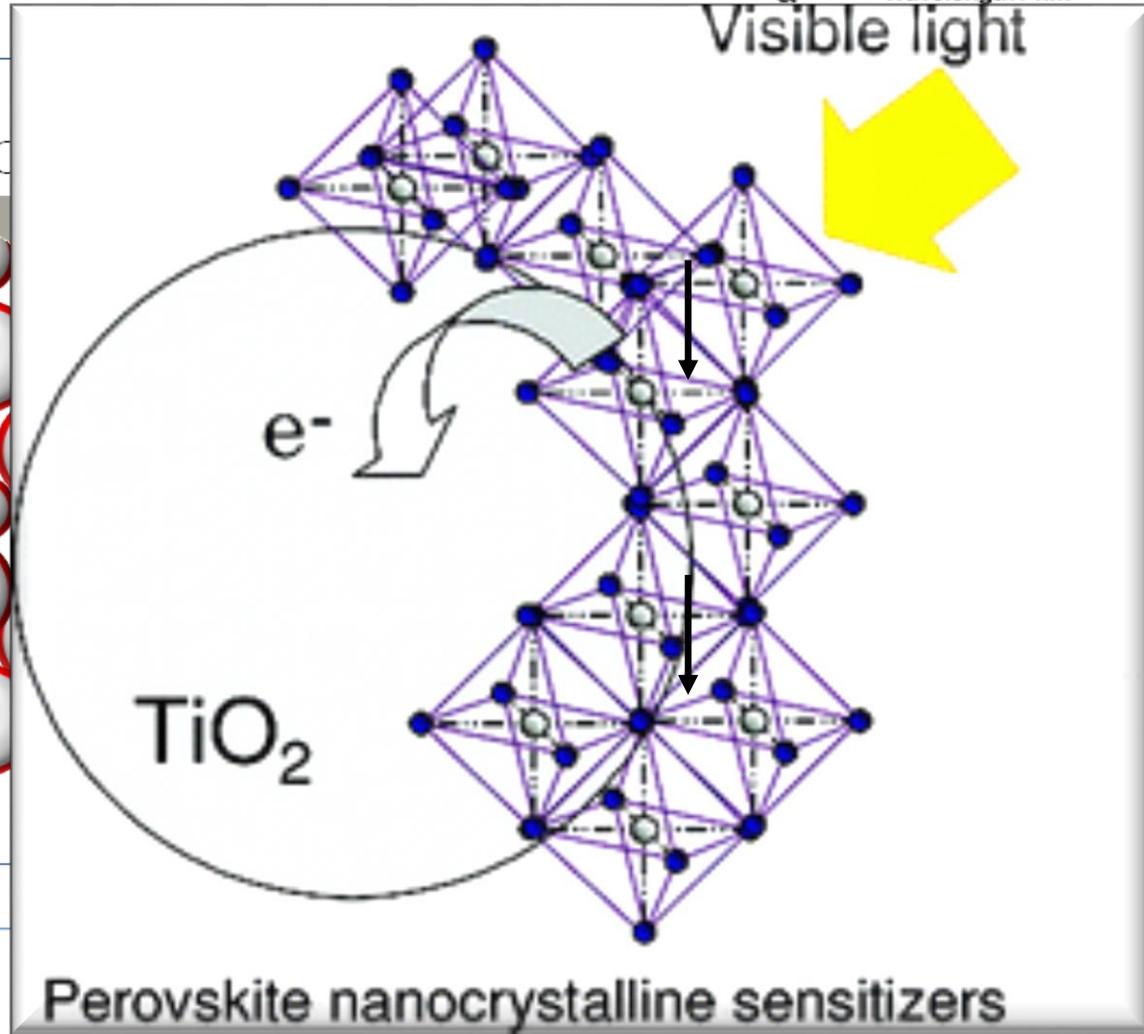
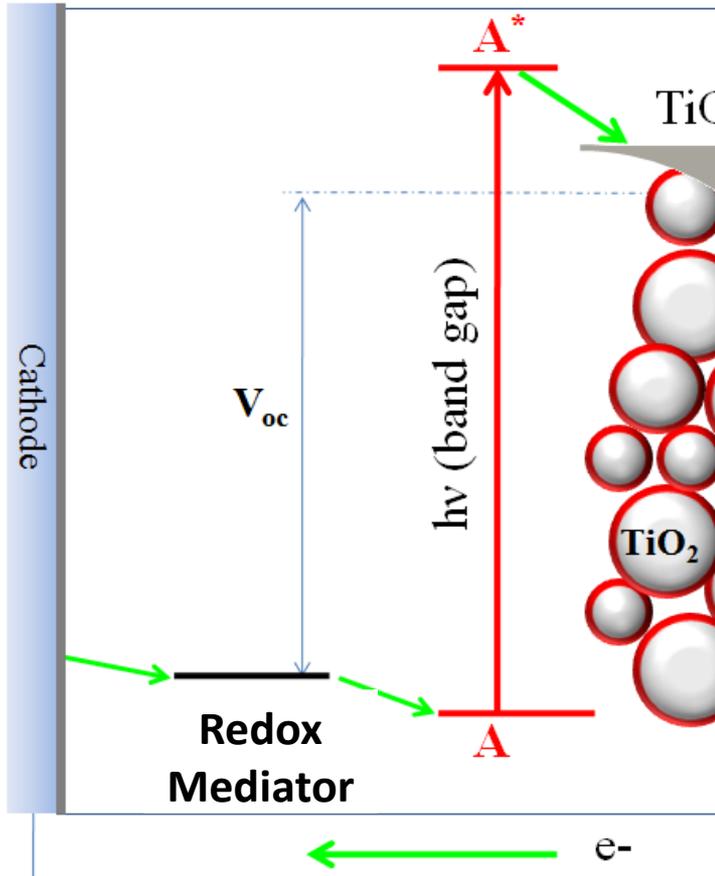
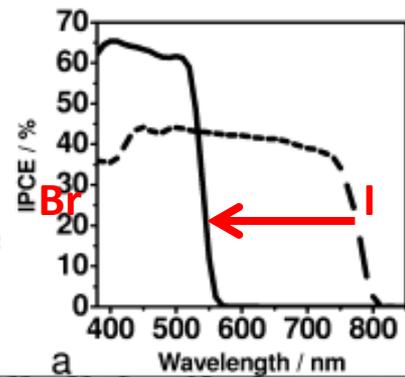
(Rev. 03-19-2015)

Perovskite solar cells: Organohalide lead perovskites



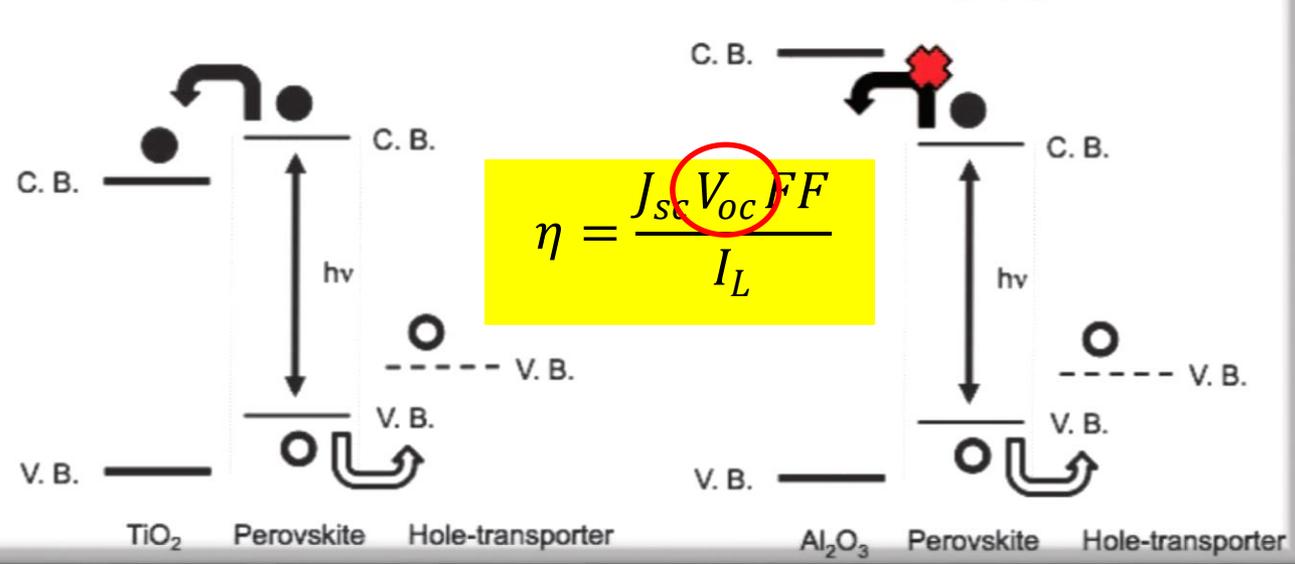
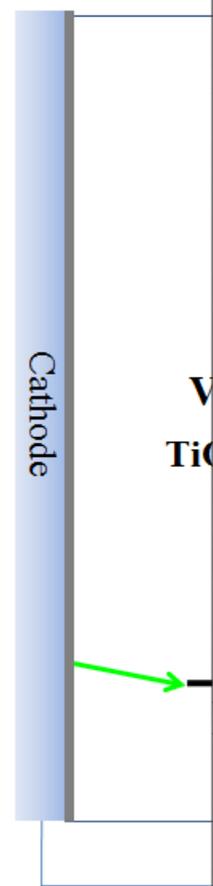
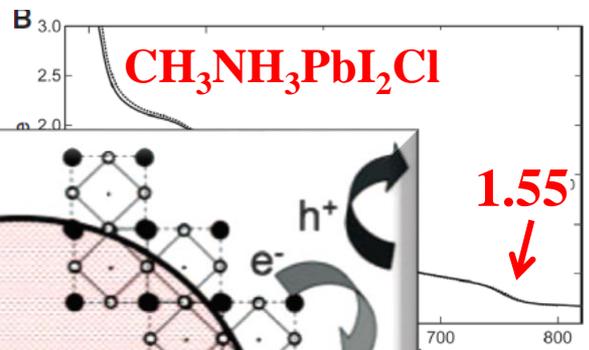
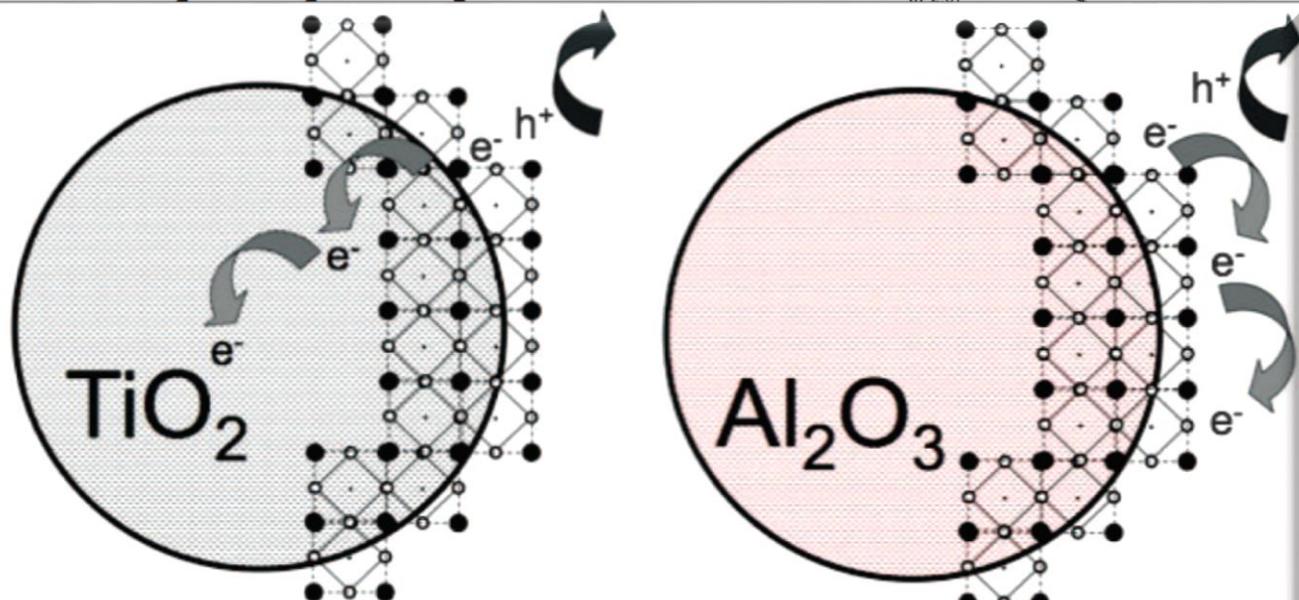
Organometal Halide Perovskites as Visible-Light Sensitizers for Photovoltaic Cells

Akihiro Kojima,[†] Kenjiro Teshima,[‡] Yasuo Shirai,[§] and Tsutomu Miyasaka^{*,†,‡,||}



Efficient Hybrid Solar Cells Based on Mesoporous TiO₂ and Al₂O₃ Organometal Halide Perovskite

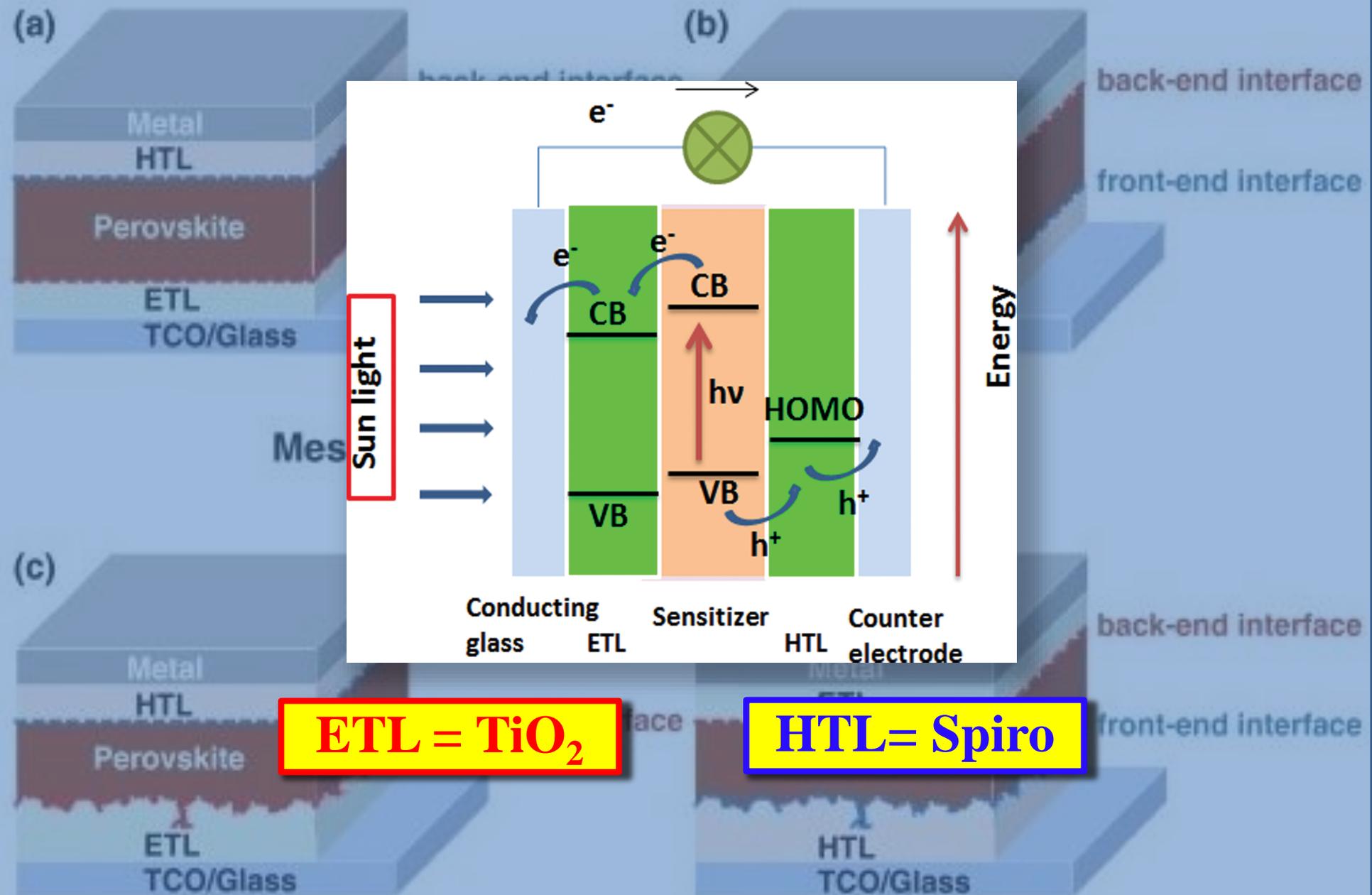
Michael M. Lee,¹



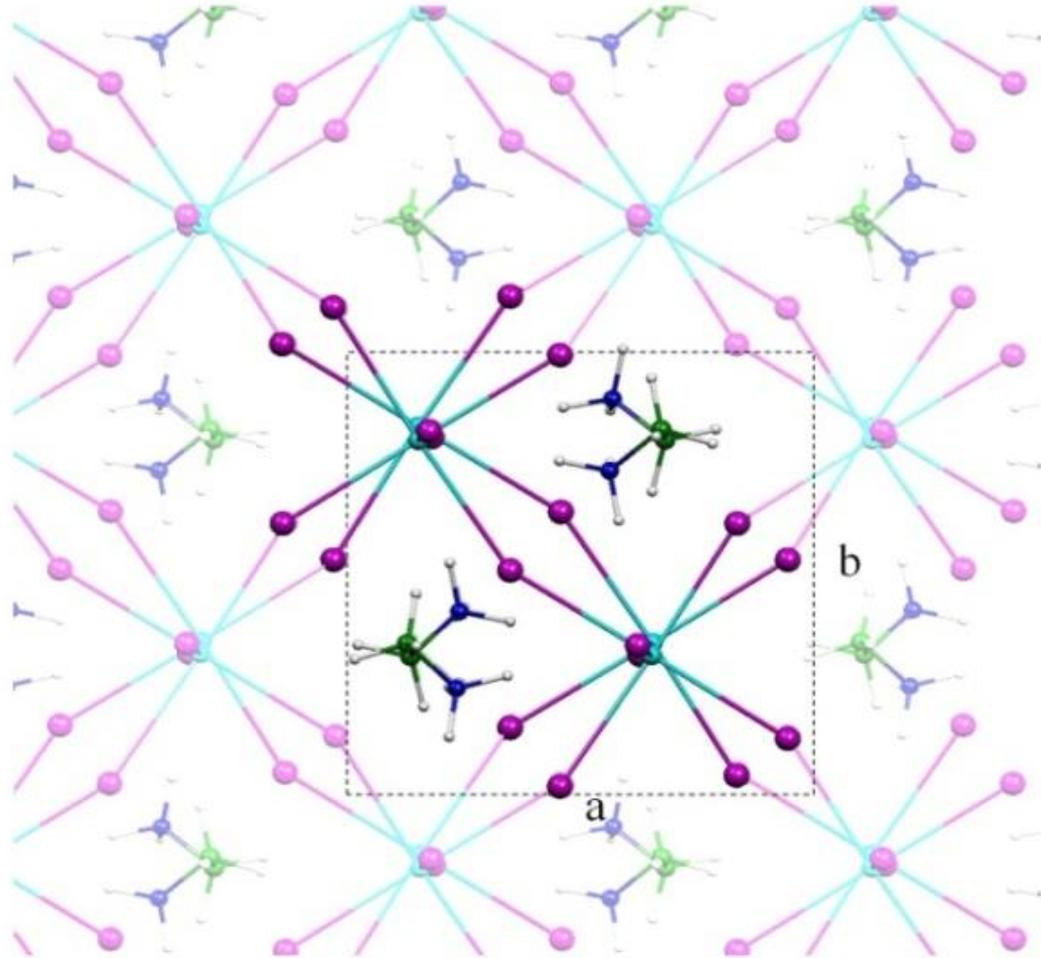
TiO₂ and
Perovskite

H.S. Kim et al. *Sci. Rep.* **2012**, 2; M. Lee et al. *Science* **2012**, 338, 643;
L. Etgar et al. *J. Am. Chem. Soc.* **2012**, 134, 17396; H. Zhou et al. *Science*, **2014**, 354, 543.

Planar structure



Tetragonal MAPbI₃: I4cm



Model:

I4cm unit cell

48 atoms

Geometry:

SR-DFT (PBE)

VdW corrections

Lattice parameters 1%

Electronic structure:

Direct band-gap at Γ

Band-gap 1.65 eV

Exp. 1.6-1.7 eV



Fortuitous agreement!!!

Spin-orbit coupling?

E. Mosconi et al. *J. Phys. Chem. C* **2013**, 114, 13092.

J. Even et al. *J. Phys. Chem. Lett.* **2013**, 114, 2999.

G. Giorgi et al. *J. Phys. Chem. Lett.* **2013**, 114, 4213.

Band-gap 0.6 eV

Perovskites Electronic Structure

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Relativistic GW calculations on $\text{CH}_3\text{NH}_3\text{PbI}_3$ and $\text{CH}_3\text{NH}_3\text{SnI}_3$ Perovskites for Solar Cell Applications

Paolo Umari^{1,2}, Edoardo Mosconi³ & Filippo De Angelis³

¹Dipartimento di Fisica e Astronomia, Università di Padova, via Marzolo 8, I-35131 Padova, Italy, ²CNR-ION DEMOCRITOS, Theory@Elettra Group, c/o Sincrotrone Trieste, Area Science Park, Basovizza, I-34012 Trieste, Italy, ³Computational Laboratory for Hybrid/Organic Photovoltaics (CLHYO), CNR-ISTM, Via Elce di Sotto 8, I-06123, Perugia, Italy.

Hybrid AMX_3 perovskites ($A = \text{Cs}, \text{CH}_3\text{NH}_3$; $M = \text{Sn}, \text{Pb}$; $X = \text{halide}$) have revolutionized the scenario of emerging photovoltaic technologies, with very recent results demonstrating 15% efficient solar cells. The $\text{CH}_3\text{NH}_3\text{PbI}_3/\text{MAPb}(\text{I}_{1-x}\text{Cl}_x)_3$ perovskites have dominated the field, while the similar $\text{CH}_3\text{NH}_3\text{SnI}_3$ has not been exploited for photovoltaic applications. Replacement of Pb by Sn would facilitate the large uptake of perovskite-based photovoltaics. Despite the extremely fast progress, the materials electronic properties which are key to the photovoltaic performance are relatively little understood. Density Functional Theory electronic structure methods have so far delivered an unbalanced description of Pb- and Sn-based perovskites. Here we develop an effective GW method incorporating spin-orbit coupling which allows us to accurately model the electronic, optical and transport properties of $\text{CH}_3\text{NH}_3\text{SnI}_3$ and $\text{CH}_3\text{NH}_3\text{PbI}_3$, opening the way to new materials design. The different $\text{CH}_3\text{NH}_3\text{SnI}_3$ and $\text{CH}_3\text{NH}_3\text{PbI}_3$ electronic properties are discussed in light of their exploitation for solar cells, and found to be dominantly due to relativistic effects. These effects stabilize the $\text{CH}_3\text{NH}_3\text{PbI}_3$ material towards oxidation, by inducing a deeper valence band edge. Relativistic effects, however, also increase the material band-gap compared to $\text{CH}_3\text{NH}_3\text{SnI}_3$, due to the valence band energy downshift (~ 0.7 eV) being only partly compensated by the conduction band downshift (~ 0.2 eV).

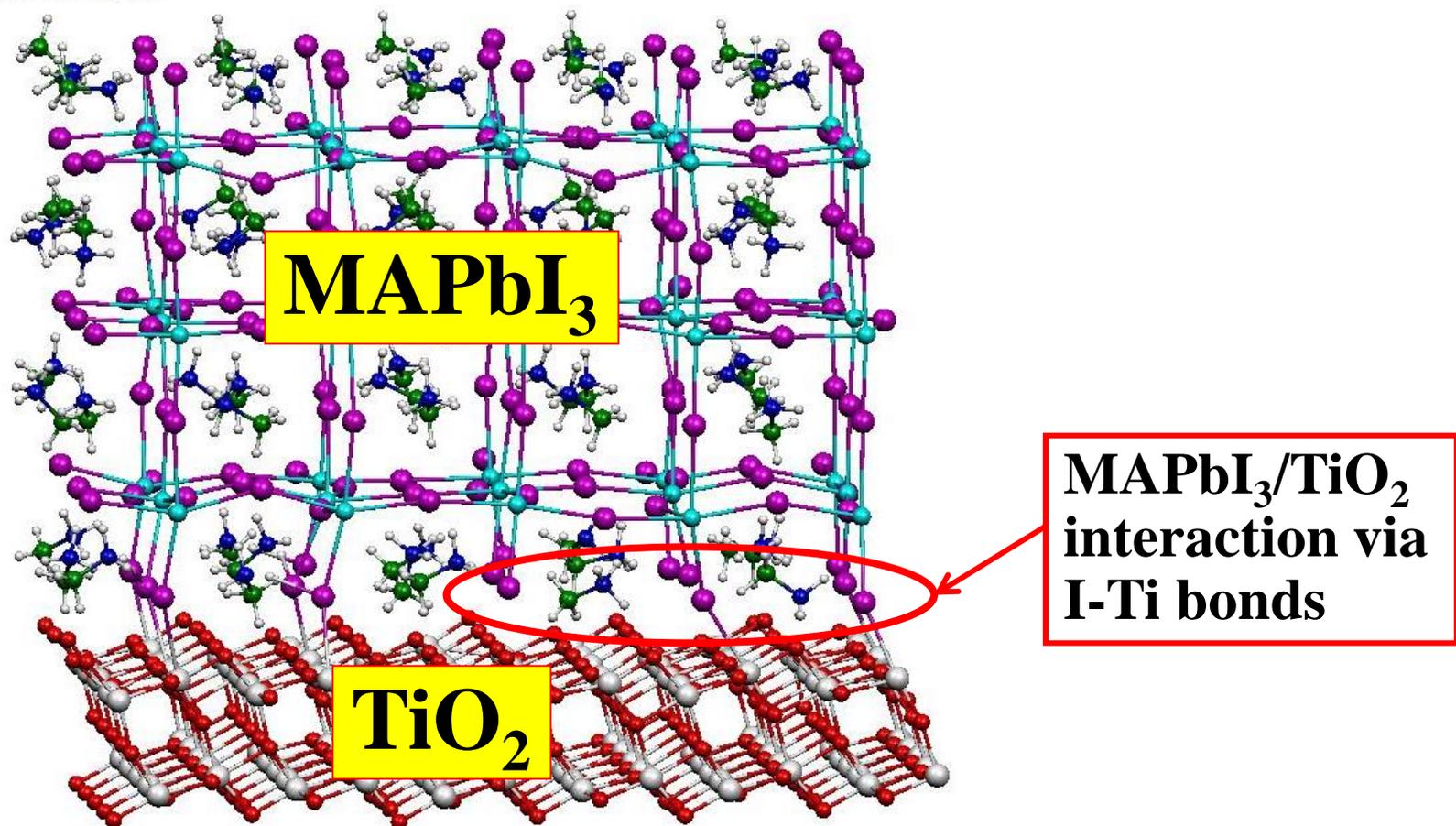
Perovskite/TiO₂ interface: Structural features

NANO LETTERS

Letter

pubs.acs.org/NanoLett

Vittoria Roiati,^{†,‡,#} Edoardo Mosconi,[§] Andrea Listorti,^{*,†,||} Silvia Colella,^{||} Giuseppe Gigli,^{†,||,⊥}
and Filippo De Angelis^{*,§}

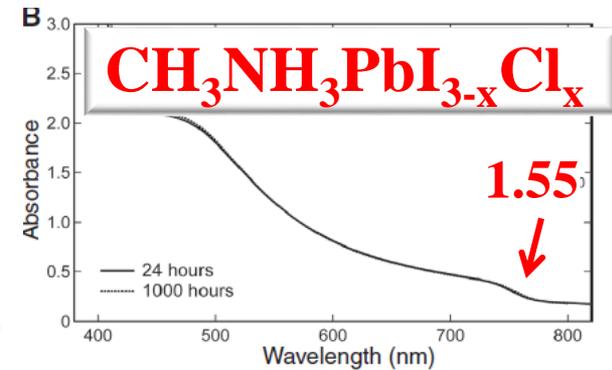


V. Roiati, E. Mosconi, et al. *Nano Lett.* **2014**, *14*, 2168.

TiO₂/perovskites interfaces: Role of Cl doping

Efficient Hybrid Solar Cells Based on Meso-Superstructured Organometal Halide Perovskites

Michael M. Lee,¹ Joël Teuscher,¹ Tsutomu Miyasaka,² Takuro N. Murakami,^{2,3} Henry J. Snaith^{1*}

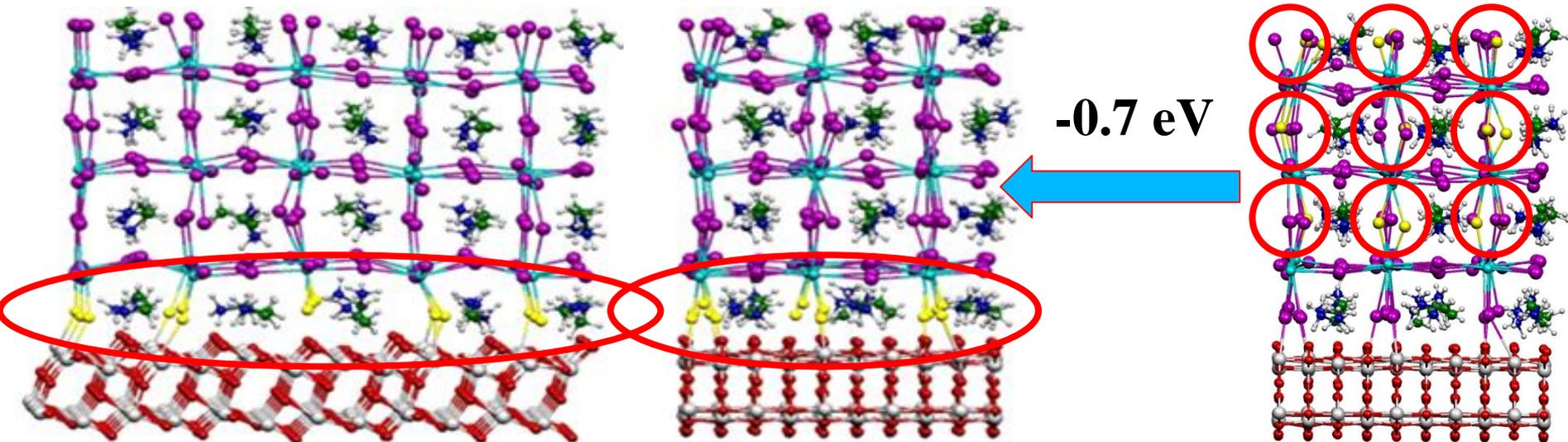


XRD \rightarrow 0.7% volume contraction



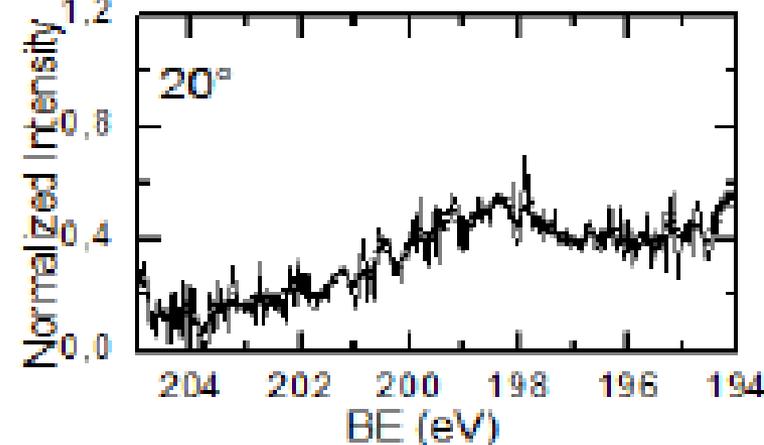
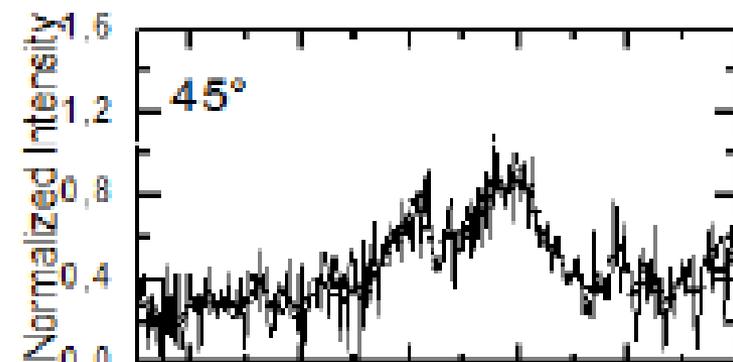
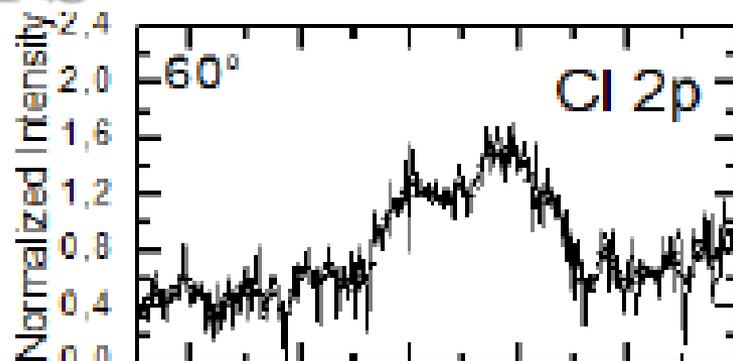
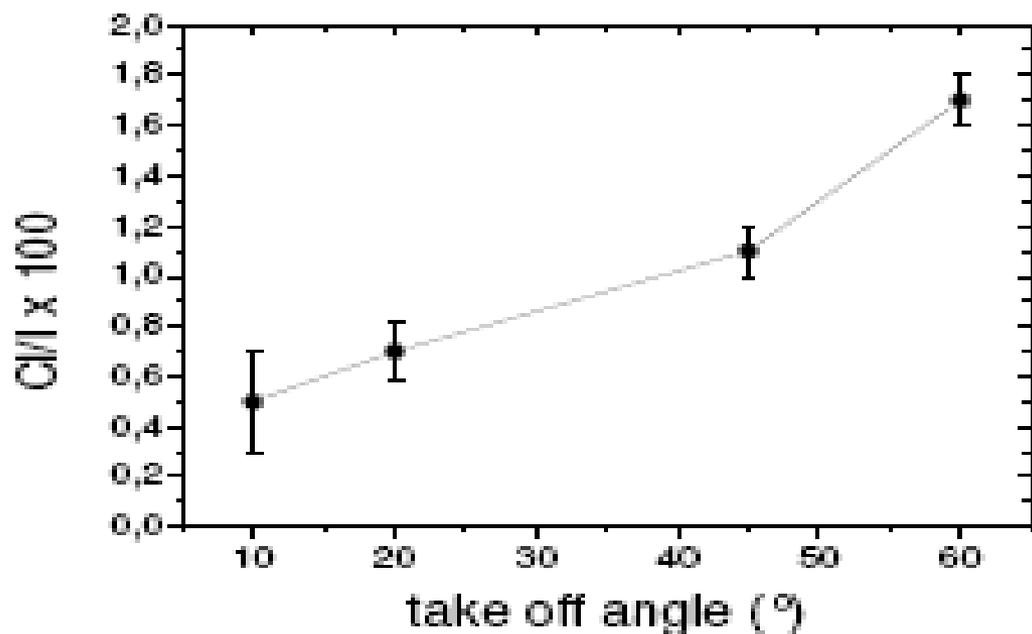
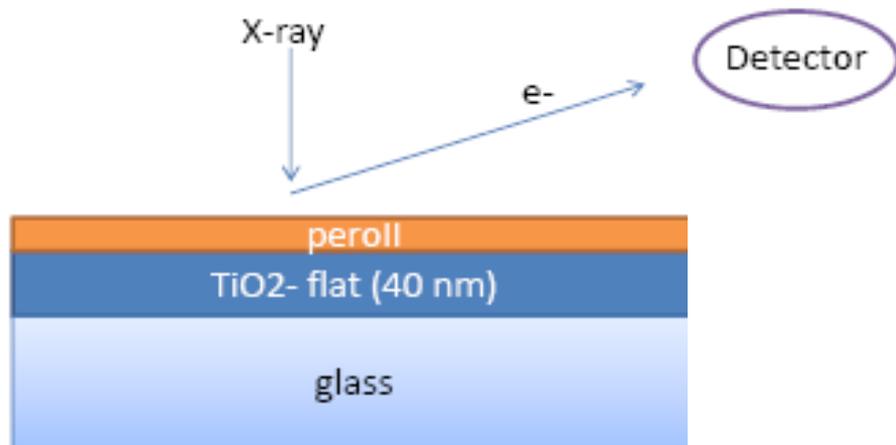
~1% Cl

S. Colella, E. Mosconi, et al. *Chem. Mater.* **2013**, 25, 4613.



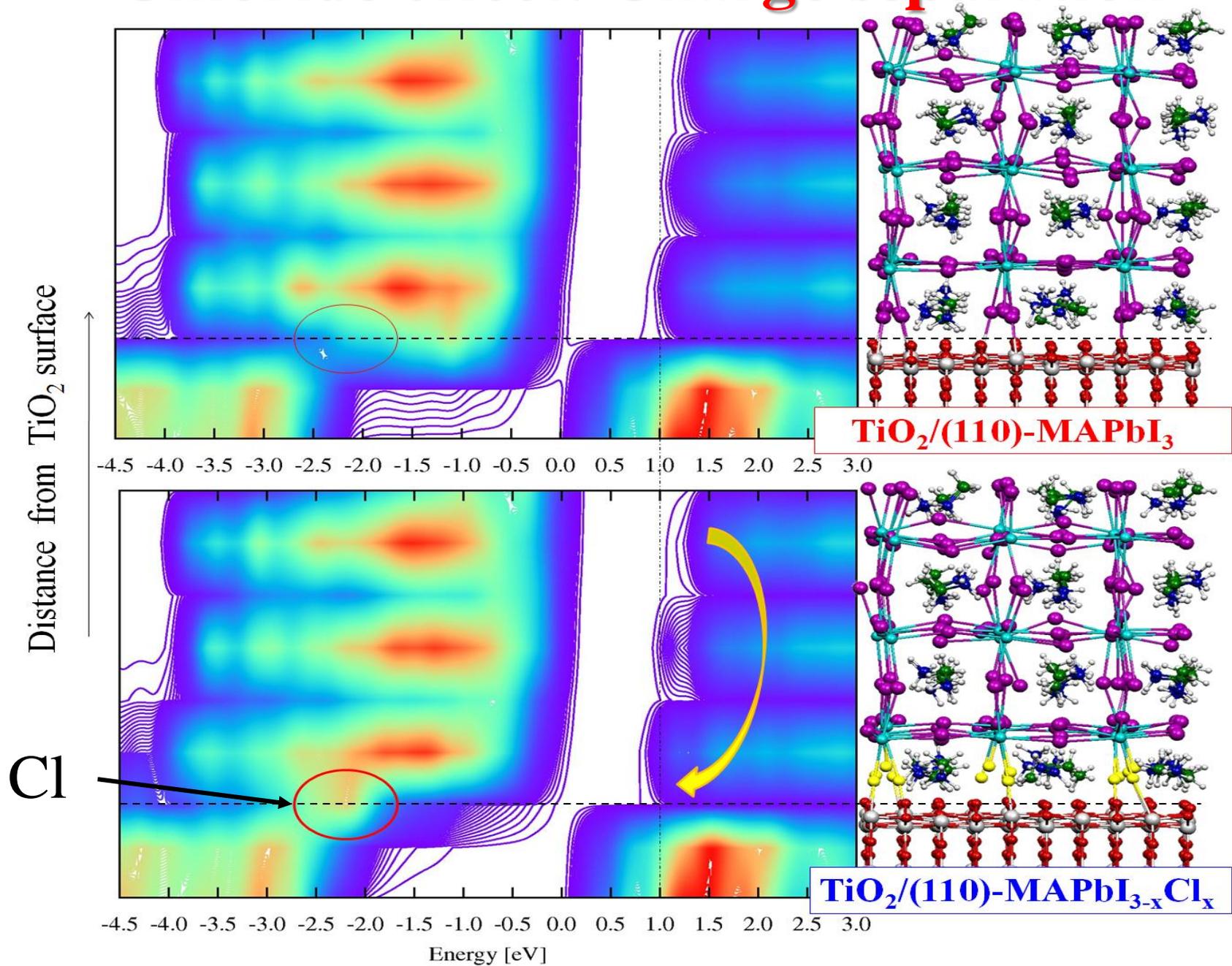
E. Mosconi, et al. *J. Phys. Chem. Lett.* **2014**, 5, 2619.

Variable angle XPS



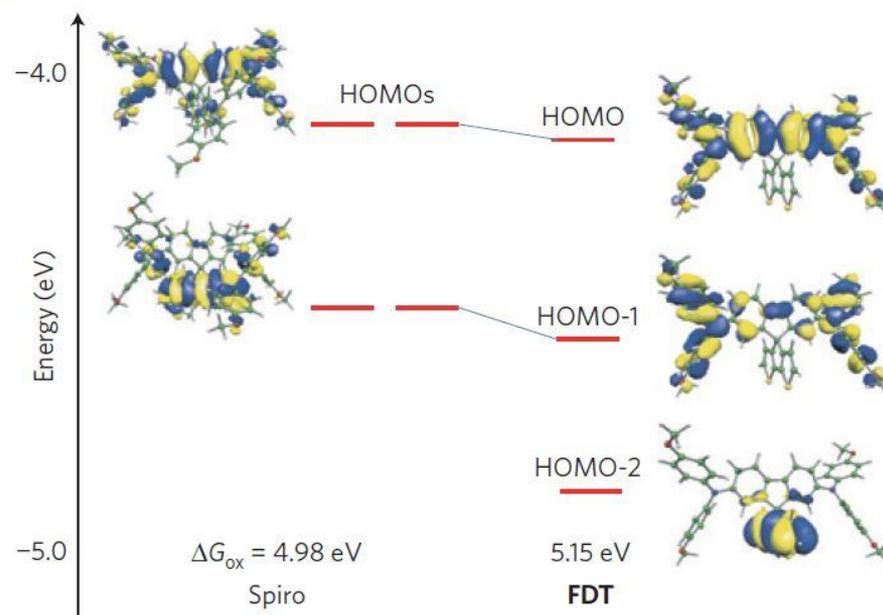
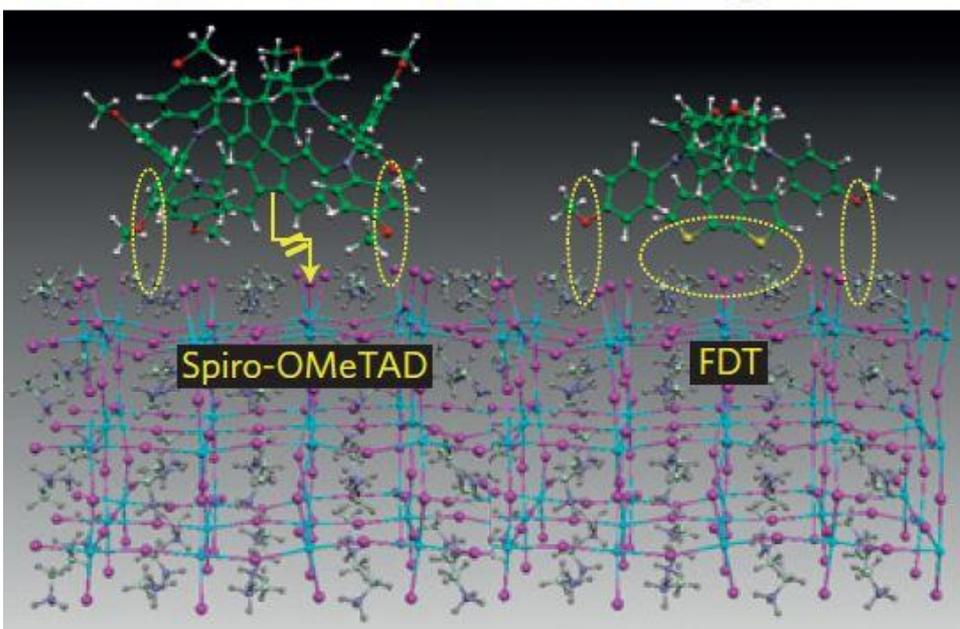
E. Mosconi, et al. *J. Phys. Chem. Lett.* **2014**, *5*, 2619.

Chloride effect: Charge separation



A molecularly engineered hole-transporting material for efficient perovskite solar cells

Michael Saliba¹, Simonetta Orlandi², Taisuke Matsui³, Sadig Aghazada¹, Marco Cavazzini², Juan-Pablo Correa-Baena⁴, Peng Gao¹, Rosario Scopelliti¹, Edoardo Mosconi⁵, Klaus-Hermann Dahmen⁶, Filippo De Angelis⁵, Antonio Abate⁷, Anders Hagfeldt⁴, Gianluca Pozzi², Michael Graetzel⁷ and Mohammad Khaja Nazeeruddin^{1*}



Ion (Defects) Migration in MAPbI₃: Activation energies

Unwanted phenomena:
Slow photo conductivity response
Hysteresis

Ions migration

Defect	E _a (eV)			
	Azpiroz - CNR	Haruyama	Eames	Meloni
I Vacancy	0.08	0.33	0.58	0.13
MA Vacancy	0.46	0.55	0.84	0.60
Pb Vacancy	1.06	-	2.31	1.39
I Interstitial	0.08	-	-	-

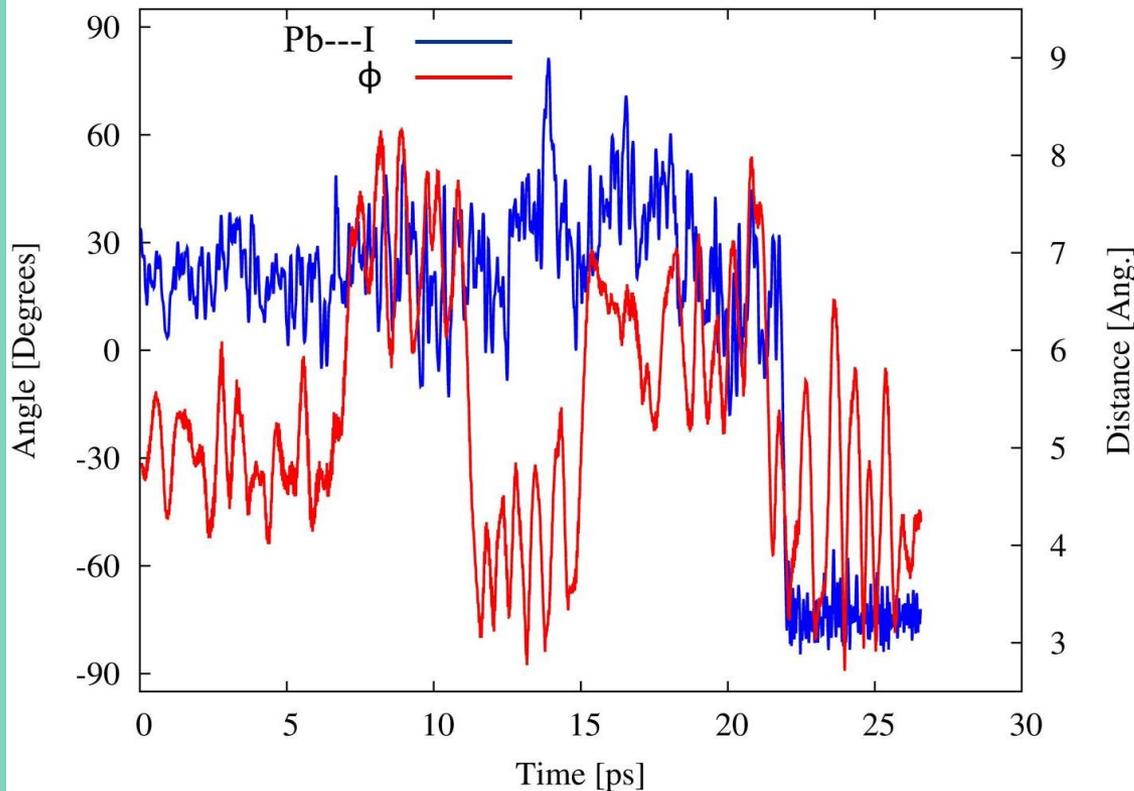
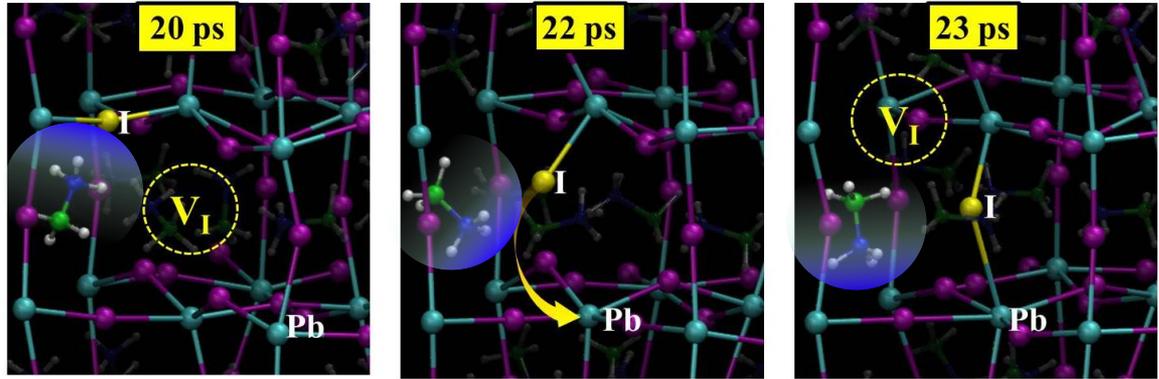
Azpiroz, J. M.; Mosconi, E.; Bisquert, J.; De Angelis, F. *Energy Environ. Sci.* 2015, 8, 2118.

Haruyama, J.; Sodeyama, K.; Han, L.; Tateyama, Y. *J. Am. Chem. Soc.* 2015, 137, 10048.

Eames, C.; Frost, J. M.; Barnes, P. R. F.; O'Regan, B. C.; Walsh, A.; Islam, M. S. *Nat Commun* 2015, 6, 7497

Meloni, S. ; Moehl, T.; Tress, ...Rothlisberger, U.; Graetzel, M. *Nat Commun* 2016, 7, 10334.

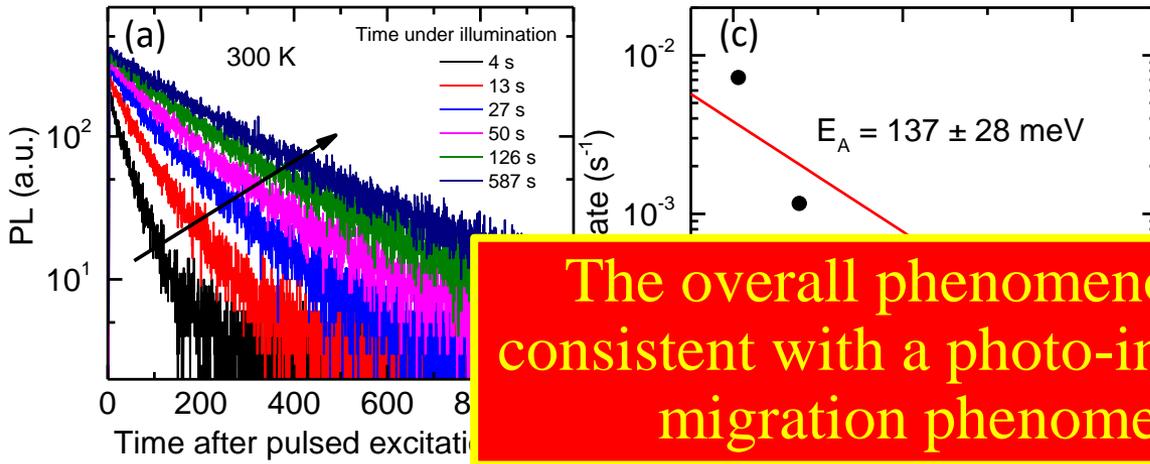
I- Vacancy Migration: Car-Parrinello MD



MA dynamics and ion/defect migration are essentially coupled:

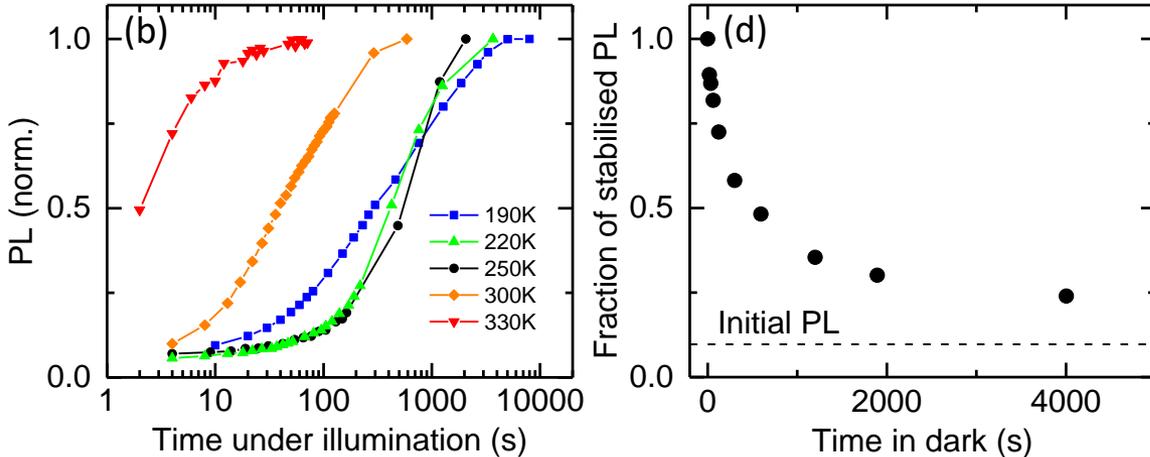
MA assists defect migration by reorienting its charge distribution along with the ionic motion, providing a local charge screening mechanism which may further speed up the ionic migration

Over a period of time under illumination, the PL lifetime and total intensity both increase and eventually stabilize, **consistent with a substantial reduction of** ... **five decay pathways**



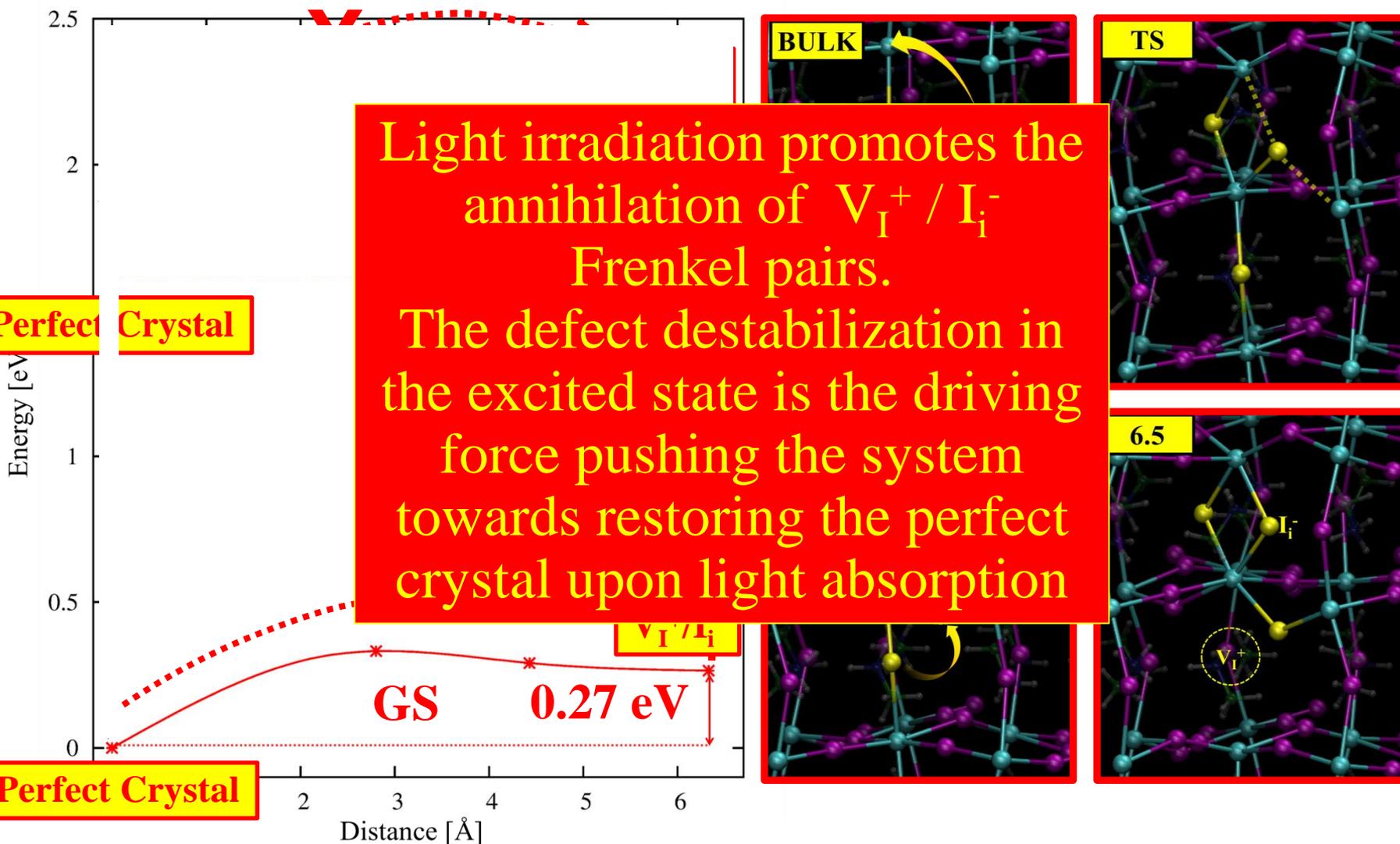
The overall phenomenology is consistent with a photo-induced ion migration phenomenon

... of trap density: from 10^{17} to $\sim 10^{16}$ cm⁻³



The observed temperature-dependent PLQE rise was fitted by an Arrhenius exponential behavior, delivering an **activation energy of 0.14 ± 0.03 eV** in line with the calculated barriers of V_I and I_i

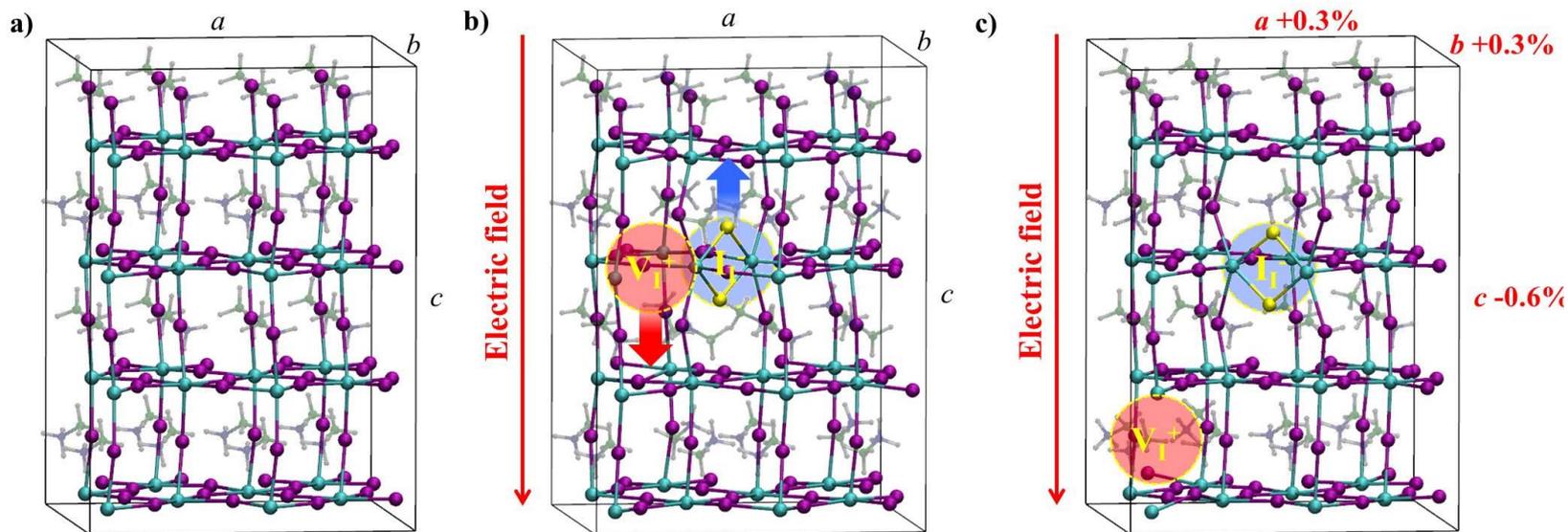
Light-induced Annihilation of Frenkel Defects in Organo-Lead Halide Perovskites



Article Published: 24 September 2018

Large electrostrictive response in lead halide perovskites

Bo Chen, Tao Li, Qingfeng Dong, Edoardo Mosconi, Jingfeng Song, Zhaolai Chen, Yehao Deng, Ye Liu, Stephen Ducharme, Alexei Gruverman, Filippo De Angelis ✉ & Jinsong Huang ✉



Conclusions and Remarks

- Electronic and Optical properties of MAPbI_3
GW-SOC, HSE06-SOC, PBE0-SOC
- Role of Cl doping
- Role of MA^+ in determining the migration mechanism
- Light-induced Annihilation of Frenkel Defects

Computational Laboratory for Hybrid/Organic Photovoltaics



Collaborations

- CNR-IMM
(Alberti, Pellegrino)
- CNR-IMEM (Mosca)
- CNR-NANO
(Colella, Listorti, Gigli)
- CNR-ISMN (Ruani)
- IIT (Petrozza)
- LPI-EPFL
(Nazeeruddin, Graetzel)
- UOXF (Snaith)

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ESPResSo
LCE-07-2016-2017-PERTPV