ARTICLE

https://doi.org/10.1038/s41467-022-29003-2

OMMUNICATIONS

OPEN



Molecular orbital theory in cavity QED environments

Rosario R. Riso ^[0], Tor S. Haugland ^[0], Enrico Ronca ^[0] ² & Henrik Koch ^[0] ^{1,3⊠}

Coupling between molecules and vacuum photon fields inside an optical cavity has proven to be an effective way to engineer molecular properties, in particular reactivity. To ease the rationalization of cavity induced effects we introduce an ab initio method leading to the first fully consistent molecular orbital theory for quantum electrodynamics environments. Our framework is non-perturbative and explains modifications of the electronic structure due to the interaction with the photon field. In this work, we show that the newly developed orbital theory can be used to predict cavity induced modifications of molecular reactivity and pinpoint classes of systems with significant cavity effects. We also investigate electronic cavityinduced modifications of reaction mechanisms in vibrational strong coupling regimes.

¹ Department of Chemistry, Norwegian University of Science and Technology, 7491 Trondheim, Norway. ² Istituto per i Processi Chimico Fisici del CNR (IPCF-CNR), Via G. Moruzzi, 1, 56124 Pisa, Italy. ³ Scuola Normale Superiore, Piazza dei Cavalieri 7, 56126 Pisa, Italy. ^{Semail: henrik.koch@sns.it}

ARTICLE

he use of strong light-matter coupling to modify molecular properties and reactivity is nowadays a very popular topic in physics and chemistry¹⁻⁴. Many groundbreaking works have shown that interaction with confined fields can impact many matter features ranging from the modification of absorption and emission spectra⁵⁻⁸ to the alteration of photochemical processes^{9,10}. In particular, the chemistry community has focused on how and to which extent reactions can be engineered by coupling with light¹¹⁻¹⁴. Interaction with quantum fields can indeed significantly affect molecular processes both in the ground state and excited states^{15,16} with reported examples of reactions slowing down¹⁴, speeding-up¹⁷ or becoming selective towards one product¹⁸. The easiest way to achieve strong coupling between light and matter is through optical cavities (See Fig. 1a), where the frequency of the electromagnetic field is determined by the geometrical features of the device¹⁹. Inside the cavity, the photonic vacuum couples to the molecular system creating polaritonic states²⁰ with distinct features²¹. Most importantly, the properties of the mixed matter-photon states can be engineered tuning the photonic part of the system, which means that

polaritons represent a very effective way to modulate matter properties in a non-invasive way²².

A detailed theoretical description of the strong coupling regime is urgently needed to develop an intuitive picture of cavity chemistry, that would also greatly ease the experimental design. However, this is a challenging task because in the strong coupling regime photons become a critical component of the quantum system and they must therefore be treated as quantum particles following quantum electrodynamics (QED). Only recently, a wider interest from the chemistry community in electron-photon systems has led to the introduction of semi empirical methods²³, variational theories²⁴ as well as many others²⁵. Only four ab initio methods have been proposed so far: QED Hartree Fock (QED-HF)²⁶, quantum electrodynamical density functional theory (QEDFT)²⁷⁻²⁹, QED coupled cluster (QED-CC)^{26,30-32} and QED full configuration interactions (QED-FCI)^{26,33}. Molecular orbital theory for the strong coupling regime has so far not been proposed, although all the methods mentioned above use an orbital basis to parametrize the wave function^{26,34,35}. The molecular orbital (MO) concept is a powerful theoretical tool used to



Fig. 1 Orbital properties in optical cavities. a Pictorial representation of an optical cavity with an injection input, the coated glasses, and a spacer to regulate the distance between the mirrors. The frequency of the cavity fields is proportional to the inverse of the distance between the two mirrors. The lambda parameter quantifies the strength of the light-matter coupling. The cavity picture is based on a video realized by the Ebbesen group. **b** Origin invariance of the SC-QED-HF orbitals compared with the origin dependence of QED-HF orbitals. If the methoxy ion is displaced in space, both the orbital energies and the orbital shapes changes for QED-HF orbitals, while the SC-QED-HF orbitals remain unaltered. **c** Orbital modifications due to changes in the cavity parameters. Changing the cavity parameters (frequency and coupling) the electronic ground state and therefore the molecular orbitals are changed. Particularly, we show that an avoided crossing like situation can be observed among orbitals.

develop correlated theories and to provide a qualitative and simple interpretation of molecular properties³⁶. In particular, since MOs also display local properties of the system like the electrophilicity or nucleophylicity of atomic fragments, reactivity can in many cases be more easily rationalized in terms of orbitals, rather than with electron densities. The introduction of a MO theory for systems in QED environments may therefore significantly enhance our chemical intuition about experiments in strong coupling regimes.

In standard quantum chemistry, molecular orbitals are obtained as solutions to the Hartree Fock (HF) equations, however a proper extension of this concept to QED environments has proven to be non-trivial. Indeed, the orbitals provided by QED-HF display unphysical behaviour for charged systems with both the orbital energies and the shapes changing when the molecule is translated (See Fig. 1b). The orbital construction scheme is therefore not reliable with critical quantities like the HOMO-LUMO gap³⁷ being ill-defined. Moreover, standard perturbation theory to describe electron-electron and electron-photon correlation cannot be used without a proper orbital theory²⁶. In this work, we introduce a new ab initio electron-photon framework called strong coupling quantum electrodynamics Hartree Fock (SC-QED-HF) which naturally leads to the first fully consistent MO theory in QED environments. Our analysis reveals that in order to obtain a well behaved molecular orbital description the photonic field contribution must be accounted for in a nonperturbative way, ensuring that in the limit of infinite coupling the exact wave function is obtained. The newly developed method captures electron-photon correlation allowing us to display field induced effects on the electronic ground state while keeping the computational cost lower than post HF approaches.

Results

The photon-matter interaction is described through the well known dipole Hamiltonian $^{38-40}$

$$H = H_e + \omega b^{\dagger} b - \lambda \sqrt{\frac{\omega}{2}} (\mathbf{d} \cdot \boldsymbol{\epsilon}) (b + b^{\dagger}) + \frac{\lambda^2}{2} (\mathbf{d} \cdot \boldsymbol{\epsilon})^2, \quad (1)$$

where H_e is the electronic Born-Oppenheimer Hamiltonian, **d** is the molecular dipole operator, $\lambda = \sqrt{\frac{1}{\epsilon_0 V}}$ is the coupling constant between the molecule and the field, where V is the quantization volume. The ω and ϵ are the frequency and polarization of the electric field, and the bosonic operators b and b[†] respectively annihilate and create photons. Besides H_e , the dipole Hamiltonian is composed of three main terms: the purely photonic term, an interaction term and the dipole self energy $\frac{\lambda^2}{2} (\mathbf{d} \cdot \boldsymbol{\epsilon})^2$. The self energy is often neglected, however this term is critical to ensure the Hamiltonian is bounded from below and displays the correct scaling with the size of the system⁴¹. To describe a molecule in a cavity, we must solve the eigenvalue problem for the dipole Hamiltonian

$$H|\psi\rangle = E|\psi\rangle,\tag{2}$$

where $|\psi\rangle$ is the electron-photon wave function. Within the SC-QED-HF approach the wave function is approximated as

$$|\psi\rangle = \exp\left(-\frac{\lambda}{\sqrt{2\omega}}\sum_{p\sigma}\eta_{p}a_{p\sigma}^{\dagger}a_{p\sigma}(b-b^{\dagger})\right)|\mathrm{HF}\rangle\otimes|0_{\mathrm{ph}}\rangle.$$
 (3)

Here $a_{p\sigma}^{\dagger}$ and $a_{p\sigma}$ respectively create and annihilate an electron in orbital p with spin σ , |HF \rangle denotes an electronic Slater determinant and $|0_{\rm ph}\rangle$ is the photonic vacuum. The electron-photon correlation basis defining the orbitals p is obtained by diagonalizing the operator ($\mathbf{d} \cdot \boldsymbol{\epsilon}$). The η_p parameters in Equation (3) are orbital specific coherent state coefficients as explained in more

details in the Supplementary Methods. Using SC-QED-HF we obtain origin invariant and size-extensive energies, and most importantly, we can construct fully origin invariant molecular orbitals, see Fig. 1b. Non size-extensive effects in the cavity can be modeled by optimizing the correlation basis and including more photons in the reference wave function in Equation (3). This modification could be crucial if long range collective effects need to be modeled. Related approaches have been proposed by Mandal et al.²⁵, Ashida et al.³⁹ and Schäfer et al.⁴². These methods also use a unitary electron-photon transformation to partially decouple the electrons and photons. In particular, the approaches in Refs. ³⁹ and⁴² also start from the solution of the infinite coupling case and we therefore expect the same eigenvalues in this limit. The method in Ref. 25, on the other hand, is based on a CI like scheme where the dipole operator is diagonalized in a limited configuration space. However, developing an MO theory based on this method is not trivial.

Importantly, since the unitary SC-QED-HF transformation in Equation (3) is variationally optimized, the description of the system for finite coupling is dramatically improved (See Supplementary Fig. 1). We emphasize that SC-QED-HF can also be used to study any kind of strongly coupled systems such as plasmonic nanocavities^{43,44} and superconducting circuits⁴⁵.

As displayed in Fig. 1c, both frequency and coupling variations can induce significant modifications of the MOs. Both λ and ω can be varied by changing the geometrical features of the cavity – opening the possibility of engineering ground state molecular properties by tuning the field parameters. We point out that while the coupling is intrinsically a light-matter property, the frequency is a field property only. The frequency dependent orbitals of SC-QED-HF are dressed with the photon field according to

$$\tilde{a}_{p\sigma}^{\dagger} = \sum_{q} a_{q\sigma}^{\dagger} \exp\left(\frac{\lambda}{\sqrt{2\omega}} \eta_{q}(b-b^{\dagger})\right) U_{qp}, \tag{4}$$

where U is the unitary transformation from the HF basis to the correlated basis. Both occupied and unoccupied orbitals are affected by the cavity parameters, however, larger changes are typically observed for the latter. Molecular orbitals are used to qualitatively predict and interpret molecular excitations as well as bond formations, therefore the cavity-induced modification of the orbital character reported in Fig. 1c suggests significant changes in the molecular properties. Orbital avoided crossings and orbital mixing are other effects that can be induced by coupling to the photon field. In particular, when two orbitals have the same symmetry, an "avoided crossing" like situation can be observed (the orbitals mix but their energies never become degenerate, see Fig. 1c). Conversely, when two almost degenerate orbitals have different symmetry, a crossing is allowed and the orbital ordering is simply exchanged, see Supplementary Fig. 2. Whenever the HOMO-LUMO gap is small we expect to see particularly large cavity effects and to achieve significant control over molecular properties. In these cases the near degeneracy between HOMO and LUMO can be enhanced or lifted by the cavity fields²⁶ with considerable impact on the ground state wave function. A reliable scheme to model these kinds of systems might require combining the SC-QED-HF Hamiltonian with an active space method⁴⁶.

One of the main applications of molecular orbital theory is the rationalization of molecular reactivity. Specifically, MOs are used to predict the critical steps of a reaction, as well as the effects that a modification of the conditions can induce on the final outcome. Here we investigate whether the molecular orbitals provided by SC-QED-HF can be used to predict cavity effects on the formation of a complex from its initial constituents. As a case study, we chose a prototypical example of a nucleophilic substitution (SN2): a methyl chloride reacting with an ammonia molecule to produce



Fig. 2 Orbital rationalization of molecular reactivity. a Orbital analysis of the reaction between ammonia and methyl chloride. Our discussion focuses on the analysis of the only occupied orbital having bonding character on both reagents, the HOMO-5 orbital. We notice that as the coupling increases the bond between carbon and nitrogen becomes weaker. In particular, for coupling $\lambda = 0.2$ the HOMO-5 is a simple composition of the non-bonding ammonia lone pair and the C-Cl sigma bond. This is verified using a projection of HOMO-5 on the orbitals of the fragments. The contributions are given in percentages. **b**, **c** Difference between electron densities computed at coupling $\lambda = 0.1/0.2$ and $\lambda = 0.0$. The blue regions of the isosurfaces show charge accumulation inside the cavity while the orange regions show charge depletion. In dark blue we plot the integral of the density differences to further display cavity induced charge reorganization^{26,33}. The polarization of the field is placed along the C-Cl bond and the cavity frequency is 13.6 eV.

methylamine and hydrogen chloride. A particularly interesting configuration along the reactive path is the one reported in Fig. 2 where both the C-Cl and the C-N bonds are not fully formed nor broken. Since this arrangement is similar to the transition state of the SN2 reaction, its cavity induced stabilizations or destabilizations play a critical role in determining a reaction speedup or slowdown. Analyzing the MOs of the complex we notice that there is only one occupied orbital with a significant bonding component over both the chloride and the nitrogen, HOMO-5. The HOMO-5 orbital of the complex is composed of the nitrogen lone pair, the C-Cl sigma bond and the three C-H sigma bonds. Projection on the orbitals of the fragments allows us to monitor field induced changes in the character of HOMO-5. Specifically, in Fig. 2 we display how HOMO-5 and its projections change for three different values of λ : 0.0, 0.1 and 0.2. Coupling values of 0.1 or 0.2 are extremely high, far superior to what is nowadays experimentally achieved for molecules interacting with a single mode of the quantum field. However, larger values are used here to approximate multimode and collective effects³³ (See methods). As the coupling increases, HOMO-5 becomes more localized on the two different fragments. Specifically, the sigma bond between the nitrogen and the carbon disappears as observed from the orbitals of the fragments. Indeed their overlap slightly decreases

with the coupling. We notice in particular that for $\lambda = 0.2$, the HOMO-5 orbital has the character of a sigma bond between the carbon and the chloride plus a non-bonding lone pair from the nitrogen. The orbital analysis therefore suggests that cavity induced effects make the configuration in Fig. 2 less favourable inside the cavity than outside. This statement is confirmed using an energy argument with the formation energy of the complex ($\Delta E(\lambda) = E_{complex}(\lambda) - E_{fragments}(\lambda)$) increasing by 5.30 kJ/mol for $\lambda = 0.1$ and 17.36 kJ/mol for $\lambda = 0.2$, compared to outside the cavity.

In panels b and c of Fig. 2, we display the difference between the ground state electron densities computed with and without the cavity for different couplings ($\Delta \rho = \rho_{\lambda} - \rho_{nocav}$, $\lambda = 0.1, 0.2$). In particular, the isosurface shows changes of the electronic density due to the cavity. To further analyze the charge readjustment we report an integration of the density difference along the field polarization in Fig. 2^{26,33}. We observe that as the coupling increases there is a significant charge depletion among both the C–Cl and the C–N bonds, as expected from the orbital analysis. This behaviour confirms the conjecture presented in Refs. ^{26,29,41} that upon interaction with the field, electrons reorganize to minimize the electronic dipole component along ϵ . We note that the electronic density redistribution observed in Fig. 2 is not induced by resonance effects between a molecular excited state and the cavity field. It is due to the interaction with the quantum field only. This shows that even in the non-resonant case, modifications of molecular properties can be expected.

This charge reorganization might actually play a role in determining cavity induced modifications of molecular reactivity. This idea was also reported by Galego et al.¹⁶ and tested using a transition state method for the strong light-matter coupling case. Strong coupling between molecular vibrations and cavity fields, vibrational strong coupling (VSC), has proven to be a particularly effective way to affect the outcome of molecular reactions^{14,17,18,47}. Although the number of experiments displaying to what extent reactions can be engineered through VSC is constantly increasing, several aspects still remain unclear. Indeed, while the formation of vibrational polaritons certainly contributes to determine the observed effects, questions still remain on the role played by the electrons and the solvent^{14,48}. Moreover, it has been suggested that an important part might be played by collective effects⁴⁹ and by dark states⁵⁰. Just recently, multiple works have suggested that the reaction slowdown in the VSC regimes can be described through nuclear effects only. Explanations use either the dynamical caging of the photon fields⁵¹ or cavity induced energy redistribution among normal modes⁵². However, in both cases electron-photon interaction was approximated preventing an accurate quantification of the cavity induced electronic effects on the chemical reactivity. Since SC-QED-HF captures some electron-photon correlation we will use it to assess the relevance of the electronic changes in VSC. We will focus on electronic effects and disregard the coupling between the fields and the nuclear vibrations. In particular, this means that the formation of the vibrational polaritons is not considered in the following analysis. We compute the potential energy surfaces (PES) at the HF and SC-QED-HF level of theory for two different reactions that are known for respectively being slowed down and catalyzed under vibrational strong coupling conditions: the deprotection of 1-phenyl-2-trimethylsilylacetylene (PTA) with tetra-n-butylammonium fluoride (TBAF)¹⁴ (Fig. 3a) and the solvolysis of para-nitrophenyl acetate (PNPA) with TBAF¹⁷ (Fig. 3b). The field polarization was aligned along the Si-C and the C-O bonds, respectively, while the frequencies of the cavity field are chosen to match the ones reported in Refs. ¹⁴ and¹⁷. A first inspection of the potential energy surfaces (See Supplementary Figs. 3, 4) reveals that for both reactions the overall shape is not changed by the cavity. Specifically, in the case of PTA-deprotection an intermediate is formed before the Si-C bond is broken (second step in Fig. 3a) and a transition state barrier must be overcome. Conversely, in the case of the PNPAsolvolysis the reaction is concerted. In Fig. 3 we show the difference between the SC-QED-HF and HF potential energy surfaces. The colour maps in Fig. 3c, d visualize the QED effects highlighting which configurations are stabilized by the cavity and therefore whether we should expect a reaction slowdown or a speedup. In particular, the dark zones are stabilized and the bright zones being destabilized upon coupling to the photons. In both the analyzed cases the reagents are more stabilized than the products pointing towards an overall slowdown of both the reactions. In the PTA-deprotection case our prediction is in agreement with what is experimentally observed, while in the case of PNPA, electronic effects would predict a slowdown instead of a speedup. Most importantly, however, for the two reactions in Fig. 3 the QED modifications of the electronic structure are so small they do not seem to affect in a significant way the shape of the PES. Stabilizations reach a maximum of 0.01 eV in the case of the PTA-deprotection and to about 0.07 eV in the PNANsolvolysis case. Moreover, we point out that rate changes in the experiments are only observed if resonance between the cavity

mode and the nuclear vibrations is achieved. Conversely, similar cavity induced modifications of the PESs are observed in a wide range of cavity frequencies (top-right corner of Fig. 3c, d). In conclusion the observed cavity induced electronic effects are not large enough to justify the experimental outcome. In particular we point out that no resonance condition is found in the electronic structure properties. We therefore conjecture that electronic effects alone are of secondary importance in the vibrational strong coupling regime and that no significant variations of the reaction mechanism or of the transition state should be expected for the reported examples. This is in disagreement with what was reported by Galego et al. Although different molecular systems were studied in Ref.¹², the discrepancy with our findings is due to the inclusion of the dipole self energy contribution in the Hamiltonian in Equation $(1)^{41}$. Summarizing, our studies show that electronic effects can be relevant under the appropriate experimental conditions (see Fig. 2, near degeneracy systems) but do not seem to play an important role in resonant VSC experiments. This is reasonable considering that nuclear properties are only indirectly related to electronic effects.

Discussion

In this work we have introduced a non-perturbative electronic structure method for the strong coupling regime, strong coupling QED-HF. The approach effectively captures part of the electronphoton correlation and scales as N^3 using an optimal implementation. Moreover, it leads to the first fully consistent molecular orbital theory for QED environments. The method is completely general and can therefore be applied to any kind of quantized field environment⁴⁴. The obtained orbitals have the correct fermionic/bosonic statistics and are dressed with the photonic field as shown in Equation (4). Furthermore, in the infinite coupling limit the SC-QED-HF wave function is an exact eigenfunction of the light-matter Hamiltonian. Similarly to standard molecular orbitals, the dressed orbitals can be used to rationalize molecular reactivity. From the considered examples we confirm that electrons interacting with the quantized field get more localized and, in this way reduce the electronic dipole in the electric field direction. The importance of cavity induced electronic effects in vibrational strong coupling can also be investigated using the new method. Our study shows that QED effects on the electronic ground state can be significant under the proper experimental settings (see Fig. 2), however they only play a secondary role for the analyzed VSC cases. Therefore a cavity induced change of the transition states in Fig. 3 is unlikely; in accordance with the conclusions reported in other works^{51,53}. Furthermore, SC-QED-HF opens the way for developing perturbative methods to capture additional electron-electron and electron-photon correlation with a lower computational cost than, for instance, QED-CCSD-1. This should in particular allow us to describe non size-extensive effects in cavities that are critical to reproduce strong coupling modifications of intermolecular interactions³³. Finally, the introduction of a molecular orbital theory opens to the possibility of developing multi-level methods to simulate strong light-matter coupling for large molecular systems and include collective effects in an ab initio framework.

Methods

The SC-QED-HF approach is a fully ab initio electron-photon correlated framework. The method has been implemented in a development version of the eT program⁵⁴. The optimization of the wave function is performed using the gradients of the energy with respect to the η_p parameters and the density matrix. The gradients are computed analytically and their explicit expressions are reported in the Supplementary Methods. The DIIS extrapolation procedure is adopted to speed up convergence⁴⁶. The electronic density is initialized as a superposition of atomic densities while the starting guess for the η_p parameters is chosen to equal the eigenvalues of $\mathbf{d} \cdot \boldsymbol{\epsilon}$. We have tested other starting values of the η_p parameters,



Fig. 3 Potential energy surfaces for reactions in the VSC regime. Schematic representations of the reaction mechanisms for **a** the deprotection of PTA using TBAF and **b** the solvolysis of PNPA. The electron movement is represented by blue arrows. **c**, **d** Difference between the potential energy surfaces for the PTA deprotection/PNPA solvolysis computed with SC-QED-HF and HF. The dark regions are stabilized by the cavity induced effects while bright regions are destabilized. In the top corners we show that the energy for the PES configurations is monotone with respect to the cavity frequency and therefore no resonance effects are observed. The coupling parameters are $\lambda = 0.025$ and $\lambda = 0.044$ respectively, while the frequencies are in resonance with the experimental Si-C and the C=O vibrational frequencies. In the insets of panel **c** and **d** the resonance features of SC-QED-HF are investigated. The yellow star indicates the resonant frequency. The QED effects have been shifted so that at resonance they are equal to zero.

however the final result is not affected by the initial guess. SC-QED-HF allows us to incorporate some electron-photon correlation in the wave function without invoking the complexity of methods like coupled cluster or full configuration interaction. In particular, the current implementation of SC-QED-HF scales as N⁵ due to the calculation of the two electron integrals in the dipole basis. Future optimization of the method, including screening of the two electron integrals, will reduce the scaling to N³. Additional theoretical details regarding SC-QED-HF are reported in the Supplementary Methods. All geometries shown in this paper have been optimized using the ORCA package⁵⁵. We used DFT-B3LYP/def2-SVP basis set to optimize the geometries for the small molecules (methoxyde, methanol) as well as for the structure in Fig. 2. For the optimization of the complex in Fig. 2 the distance between the nitrogen and the carbon was fixed to 2.1 Å while the positions of all the other atoms were reoptimized. The fragments in Fig. 2 were obtained from the geometry of the complex without any additional optimization. In this way we isolated cavity induced effects neglecting modifications of the orbitals induced by the geometry relaxation. Finally, the complex formation energy ΔE was computed from the energies of the fragments at infinite separation. The geometries for the PNPA potential energy surfaces were optimized using DFT-PBE0/def2-SVP basis set while the PTA geometries were computed using DFT-B3LYP/6-31G* basis set similarly to what has already been reported in literature⁵². All the calculations reported in this paper have been performed using a cc-pVDZ basis set.

New experimental techniques keep pushing the boundaries of the ultrastrong light-matter coupling regime by extreme confinement of the field quantization volume, using for instance nanoplasmonic picocavities⁴³, where quantization volumes can be reduced down to the nm³ limit. However, some of the calculations

presented in this work have been performed for values of the light-matter coupling that can seem unrealistically high, with effective quantization volumes of around 0.2 nm³. This choice is justified because inside optical cavities molecules interact with more than one mode at a time and field induced effects from different modes sum with each other. The net effect can be approximated using a larger value of a single mode coupling to the molecule, λ_{eff} . In particular, in the QED-HF case, the multimode calculation is identical to using a single mode with an effective coupling, $\lambda_{eff} = \sqrt{N_{modes}}\lambda^{12,33}$. This relation does not hold exactly for SC-QED-HF because of the electron-photon correlation, however the net effect of a multimode approach still remains an increase of the light-matter coupling.

Data availability

The data that support the findings of this study are available from the corresponding author upon request.

Code availability

The custom code used for this study is available from the corresponding author upon request. A publicly available version of the code will be released following the time schedule of the eT program "https://etprogram.org/".

Received: 9 December 2021; Accepted: 23 February 2022; Published online: 15 March 2022

References

- Scholes, G. D. Polaritons and excitons: Hamiltonian design for enhanced coherence. Proc. R. Soc. Lon. Ser.-A 476, 20200278 (2020).
- Wilson-Rae, I. & Imamoğlu, A. Quantum dot cavity-qed in the presence of strong electron-phonon interactions. *Phys. Rev. B* 65, 235311 (2002).
- Hagenmüller, D., Schachenmayer, J., Schütz, S., Genes, C. & Pupillo, G. Cavity-enhanced transport of charge. *Phys. Rev. Lett.* 119, 223601 (2017).
- Wang, X., Ronca, E. & Sentef, M. A. Cavity quantum electrodynamical chern insulator: towards light-induced quantized anomalous hall effect in graphene. *Phys. Rev. B* 99, 235156 (2019).
- Herrera, F. & Spano, F. C. Absorption and photoluminescence in organic cavity qed. *Phys. Rev. A* 95, 053867 (2017).
- Wang, S., Scholes, G. D. & Hsu, L.-Y. Coherent-to-incoherent transition of molecular fluorescence controlled by surface plasmon polaritons. *J. Phys. Chem. Lett.* 11, 5948 (2020).
- Li, T. E., Nitzan, A. & Subotnik, J. E. Cavity molecular dynamics simulations of vibrational polariton-enhanced molecular nonlinear absorption. *J. Chem. Phys.* 154, 094124 (2021).
- DelPo, C. A. et al. Polariton transitions in femtosecond transient absorption studies of ultrastrong light-molecule coupling. J. Phys. Chem. Lett. 11, 2667 (2020).
- Hutchison, J. A., Schwartz, T., Genet, C., Devaux, E. & Ebbesen, T. W. Modifying chemical landscapes by coupling to vacuum fields. *Angew. Chem. Int. Ed.* 51, 1592 (2012).
- Mandal, A. & Huo, P. Investigating new reactivities enabled by polariton photochemistry. J. Phys. Chem. Lett. 10, 5519 (2019).
- Bennett, K., Kowalewski, M. & Mukamel, S. Novel photochemistry of molecular polaritons in optical cavities. *Faraday Discuss.* 194, 259 (2016).
- Galego, J., Climent, C., Garcia-Vidal, F. J. & Feist, J. Cavity casimir-polder forces and their effects in ground-state chemical reactivity. *Phys. Rev. X* 9, 021057 (2019).
- 13. Shalabney, A. et al. Coherent coupling of molecular resonators with a microcavity mode. *Nat. Commun.* **6**, 1 (2015).
- 14. Thomas, A. et al. Ground-state chemical reactivity under vibrational coupling to the vacuum electromagnetic field. *Angew. Chem. Int. Ed.* **128**, 11634 (2016).
- 15. Sau, A. et al. Modifying woodward-hoffmann stereoselectivity under vibrational strong coupling. *Angew. Chem. Int. Ed.* **133**, 5776 (2021).
- Galego, J., Garcia-Vidal, F. J. & Feist, J. Many-molecule reaction triggered by a single photon in polaritonic chemistry. *Phys. Rev. Lett.* 119, 136001 (2017).
- Lather, J., Bhatt, P., Thomas, A., Ebbesen, T. W. & George, J. Cavity catalysis by cooperative vibrational strong coupling of reactant and solvent molecules. *Angew. Chem. Int. Ed.* 131, 10745 (2019).
- 18. Thomas, A. et al. Tilting a ground-state reactivity landscape by vibrational strong coupling. *Science* **363**, 615 (2019).
- Siegman, A. E. Laser beams and resonators: beyond the 1960s. IEEE J. Sel. Top. Quant. 6, 1389 (2000).
- Wang, D. S., Neuman, T., Flick, J. & Narang, P. Light-matter interaction of a molecule in a dissipative cavity from first principles. *J. Chem. Phys.* 154, 104109 (2021).
- 21. Herrera, F. & Spano, F. C. Cavity-controlled chemistry in molecular ensembles. *Phys. Rev. Lett.* **116**, 238301 (2016).
- 22. Yuen-Zhou, J. & Menon, V. M. Polariton chemistry: thinking inside the (photon) box. *P. Natl Acad. Sci. USA* **116**, 5214 (2019).
- Fregoni, J., Granucci, G., Coccia, E., Persico, M. & Corni, S. Manipulating azobenzene photoisomerization through strong light-molecule coupling. *Nat. Commun.* 9, 1 (2018).
- 24. Rivera, N., Flick, J. & Narang, P. Variational theory of nonrelativistic quantum electrodynamics. *Phys. Rev. Lett.* **122**, 193603 (2019).
- Mandal, A., Montillo Vega, S. & Huo, P. Polarized fock states and the dynamical casimir effect in molecular cavity quantum electrodynamics. J. Phys. Chem. Lett. 11, 9215 (2020).
- Haugland, T. S., Ronca, E., Kjønstad, E. F., Rubio, A. & Koch, H. Coupled cluster theory for molecular polaritons: changing ground and excited states. *Phys. Rev. X* 10, 041043 (2020).
- Flick, J., Ruggenthaler, M., Appel, H. & Rubio, A. Atoms and molecules in cavities, from weak to strong coupling in quantum-electrodynamics (qed) chemistry. P. Natl Acad. Sci. USA 114, 3026 (2017).
- Ruggenthaler, M. et al. Quantum-electrodynamical density-functional theory: bridging quantum optics and electronic-structure theory. *Phys. Rev. A* 90, 012508 (2014).
- Flick, J., Schäfer, C., Ruggenthaler, M., Appel, H. & Rubio, A. Ab initio optimized effective potentials for real molecules in optical cavities: Photon contributions to the molecular ground state. ACS photonics 5, 992 (2018).
- Pavošević, F., Hammes-Schiffer, S., Rubio, A. & Flick, J. Cavity-modulated proton transfer reactions, arXiv preprint arXiv:2112.02138 (2021).
- Pavošević, F. & Flick, J. Polaritonic unitary coupled cluster for quantum computations. J. Phys. Chem. Lett. 12, 9100–9107 (2021).

- DePrince III, A. E. Cavity-modulated ionization potentials and electron affinities from quantum electrodynamics coupled-cluster theory. J. Chem. Phys. 154, 094112 (2021).
- Haugland, T. S., Schäfer, C., Ronca, E., Rubio, A. & Koch, H. Intermolecular interactions in optical cavities: an ab initio qed study. *J. Chem. Phys.* 154, 094113 (2021).
- Buchholz, F., Theophilou, I., Giesbertz, K. J., Ruggenthaler, M. & Rubio, A. Light-matter hybrid-orbital-based first-principles methods: The influence of polariton statistics. *J. Chem. Theory Comput.* 16, 5601 (2020).
- Nielsen, S. E. B, Schäfer, C., Ruggenthaler, M. & Rubio, A. Dressed-orbital approach to cavity quantum electrodynamics and beyond, arXiv:1812.00388 (2018).
- Fujimoto, H. & Fukui, K. In Adv. Quantum Chem., vol. 6 (Elsevier, 1972), pp. 177–201.
- Teunissen, J. L., De Proft, F. & De Vleeschouwer, F. Tuning the homo-lumo energy gap of small diamondoids using inverse molecular design. J. Chem. Theory Comput. 13, 1351 (2017).
- Di Stefano, O. et al. Resolution of gauge ambiguities in ultrastrong-coupling cavity quantum electrodynamics. *Nat. Phys.* 15, 803 (2019).
- Ashida, Y., İmamoğlu, A. & Demler, E. Cavity quantum electrodynamics at arbitrary light-matter coupling strengths. *Phys. Rev. Lett.* 126, 153603 (2021).
- Taylor, M. A., Mandal, A., Zhou, W. & Huo, P. Resolution of gauge ambiguities in molecular cavity quantum electrodynamics. *Phys. Rev. Lett.* 125, 123602 (2020).
- Rokaj, V., Welakuh, D. M., Ruggenthaler, M. & Rubio, A. Light-matter interaction in the long-wavelength limit: no ground-state without dipole selfenergy. J. Phys. B-. Mol. Opt. 51, 034005 (2018).
- Schäfer, C., Buchholz, F., Penz, M., Ruggenthaler, M. & Rubio, A. Making ab initio qed functional(s): Nonperturbative and photon-free effective frameworks for strong light-matter coupling. *Proc. Natl. Acad. Sci. USA* 118, e2110464118 (2021).
- Chikkaraddy, R. et al. Single-molecule strong coupling at room temperature in plasmonic nanocavities. *Nature* 535, 127 (2016).
- 44. Fregoni, J. et al. Strong coupling between localized surface plasmons and molecules by coupled cluster theory. *Nano Lett.* **21**, 6664 (2021).
- Gu, X., Huai, S.-N., Nori, F. & Liu, Y.-X. Polariton states in circuit qed for electromagnetically induced transparency. *Phys. Rev. A* 93, 063827 (2016).
- Helgaker, T., Jorgensen, P. & Olsen, J. Molecular electronic-structure theory (John Wiley & Sons, 2014).
- George, J., Lather, J., Thabassum, A. N. K. & Singh, J. Cavity catalysis: modifying linear free-energy relationship under cooperative vibrational strong coupling, *ChemRxiv* (2021).
- Climent, C. & Feist, J. On the sn 2 reactions modified in vibrational strong coupling experiments: reaction mechanisms and vibrational mode assignments. *Phys. Chem. Chem. Phys.* 22, 23545 (2020).
- Sidler, D., Schäfer, C., Ruggenthaler, M. & Rubio, A. Polaritonic chemistry: collective strong coupling implies strong local modification of chemical properties. J. Phys. Chem. Lett. 12, 508 (2020).
- Du, M. & Yuen-Zhou, J. Can dark states explain vibropolaritonic chemistry?, arXiv:2104.07214 (2021).
- Li, X., Mandal, A. & Huo, P. Theory of mode-selective chemistry through polaritonic vibrational strong coupling. *J. Phys. Chem. Lett.* 12, 6974 (2021).
- Schäfer, C., Flick, J., Ronca, E., Narang, P. & Rubio, A. Shining light on the microscopic resonant mechanism responsible for cavity-mediated chemical reactivity, arXiv:2104.12429 (2021).
- Li, T. E., Nitzan, A. & Subotnik, J. E. On the origin of ground-state vacuum-field catalysis: equilibrium consideration. *J. Chem. Phys.* 152, 234107 (2020).
- Folkestad, S. D. et al. et 1.0: An open source electronic structure program with emphasis on coupled cluster and multilevel methods. *J. Chem. Phys.* 152, 184103 (2020).
- 55. Neese, F. The orca program system. Wires Comput. Mol. Sci. 2, 73 (2012).

Acknowledgements

We acknowledge Matteo Castagnola and Sander Roet for insightful discussions. R.R.R., T.S.H. and H.K. acknowledge funding from the Research Council of Norway through FRINATEK Project Nos. 263110 and 275506 and the Marie Skłodowska-Curie European Training Network "COSINE– COmputational Spectroscopy In Natural sciences and Engineering," Grant Agreement No. 765739. We acknowledge computing resources through UNINETT Sigma2–the National Infrastructure for High Performance Computing and Data Storage in Norway, through Project No. NN2962k. This work has received funding from the European Research Council (ERC) under the European Union's Horizon 2020 Research and Innovation Programme (grant agreement No. 101020016).

ARTICLE

Author contributions

R.R.R., T.S.H. and H.K. developed the theory. H.K. conceived the project. R.R.R. implemented the computational methodology in the eT program. R.R.R. and E.R. proposed the investigated applications. R.R.R., T.S.H. and E.R. obtained the data. R.R.R prepared a first draft of the paper. E.R. and R.R.R. worked at the graphical part of the manuscript. H.K. and E.R. oversaw the project. All the authors discussed the results and edited the manuscript.

Competing interests

The authors declare no competing interests.

Additional information

Supplementary information The online version contains supplementary material available at https://doi.org/10.1038/s41467-022-29003-2.

Correspondence and requests for materials should be addressed to Henrik Koch.

Peer review information *Nature Communications* thanks the anonymous reviewers for their contribution to the peer review of this work.

Reprints and permission information is available at http://www.nature.com/reprints

Publisher's note Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.

Open Access This article is licensed under a Creative Commons Attribution 4.0 International License, which permits use, sharing, adaptation, distribution and reproduction in any medium or format, as long as you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made. The images or other third party material in this article are included in the article's Creative Commons license, unless indicated otherwise in a credit line to the material. If material is not included in the article's Creative Commons license and your intended use is not permitted by statutory regulation or exceeds the permitted use, you will need to obtain permission directly from the copyright holder. To view a copy of this license, visit http://creativecommons.org/ licenses/by/4.0/.

© The Author(s) 2022